



Application 1

Plastic recycling

Secondary fuels in the cement industry



Challenge

- Not sorted by type, foreign substances (metals, paper, org. Material)
- Fluctuating material flows complicate stable processes
- Faulty workmanship: damage, quality and environmental problems

XRF Solutions

- Rapid screening of raw materials
- Identification of critical components
- Contamination testing of the final product
- Standard-less semi-quantitative analysis



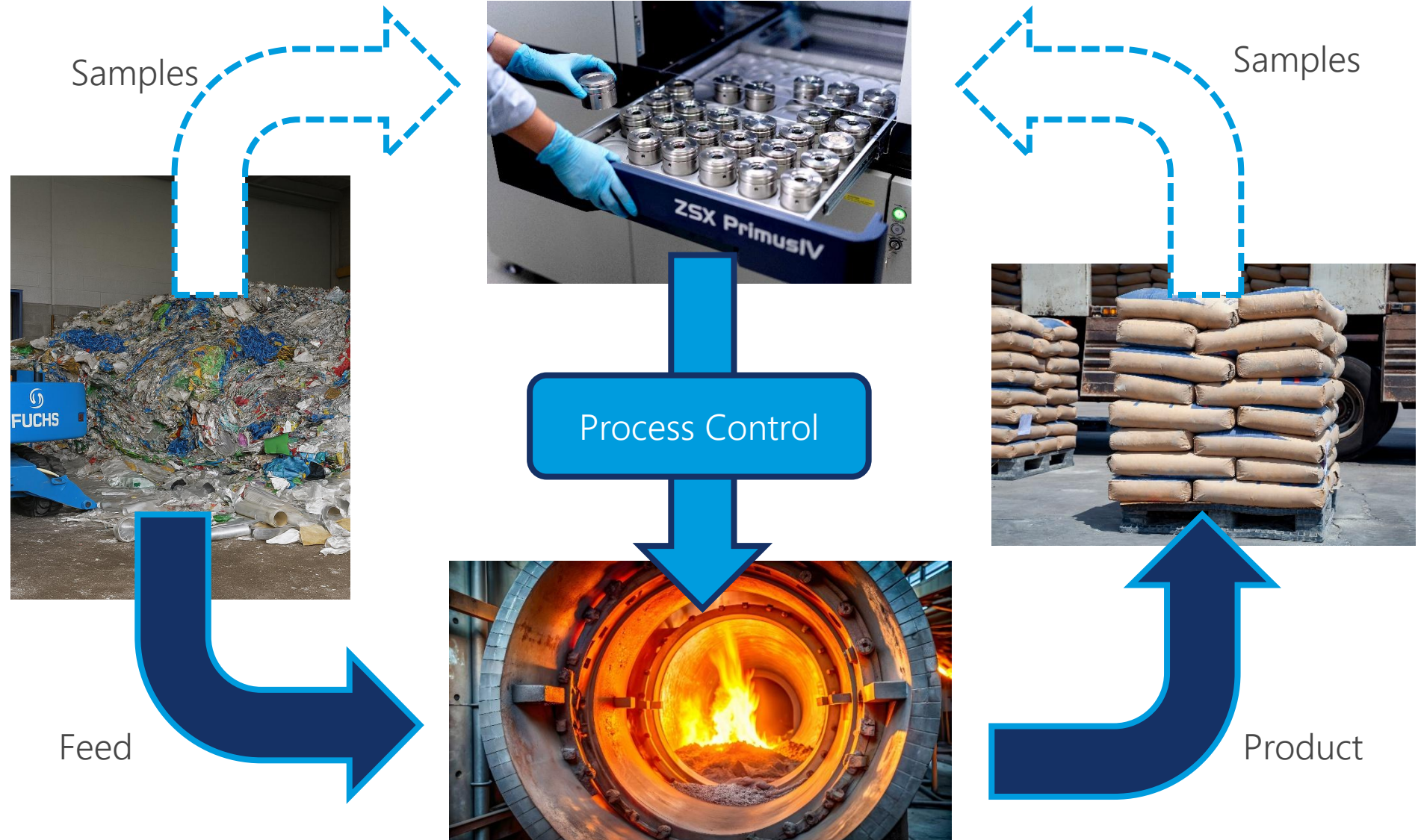
Secondary fuels in the cement industry

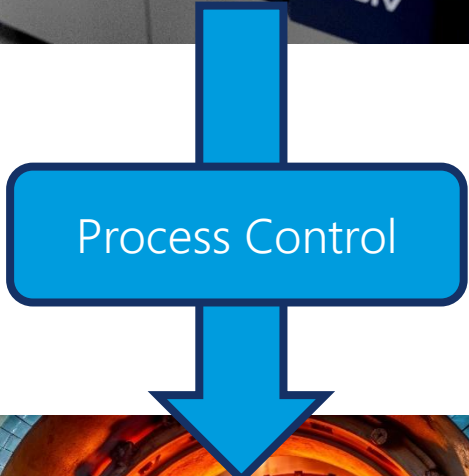
- Substitution of fossil fuels
- Cost reduction
- CO₂ reduction (from fossil fuels)



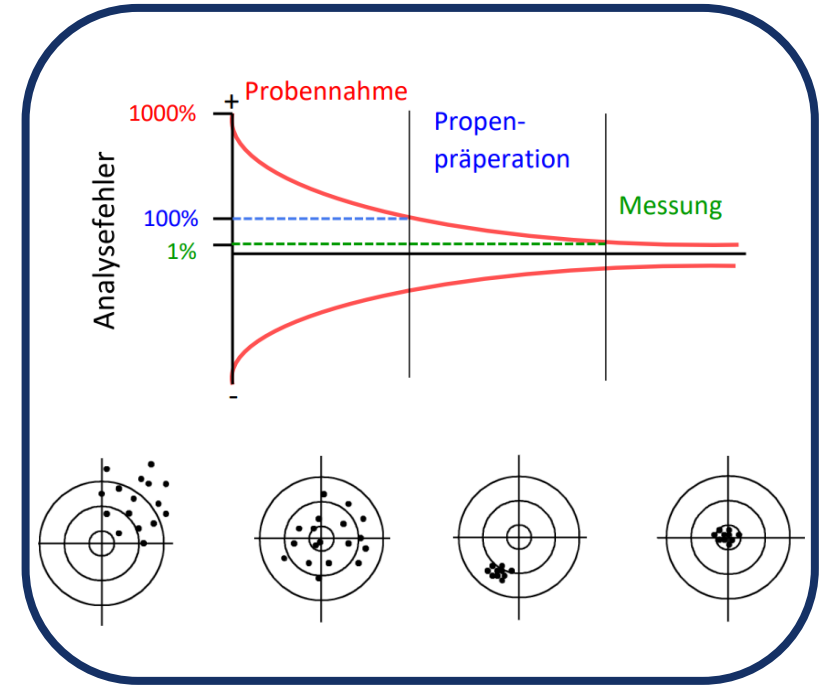
Elements to Monitor

- **P:** Severe impairment of strength made of flame retardants in plastic
- **V, Zn, Pb:** Increase the reactivity of the clinker Additives such as zinc stearates
- **Mn, Ti:** Interfere with the hydration properties of MnO/TiO₂ as pigments or UV stabilizers

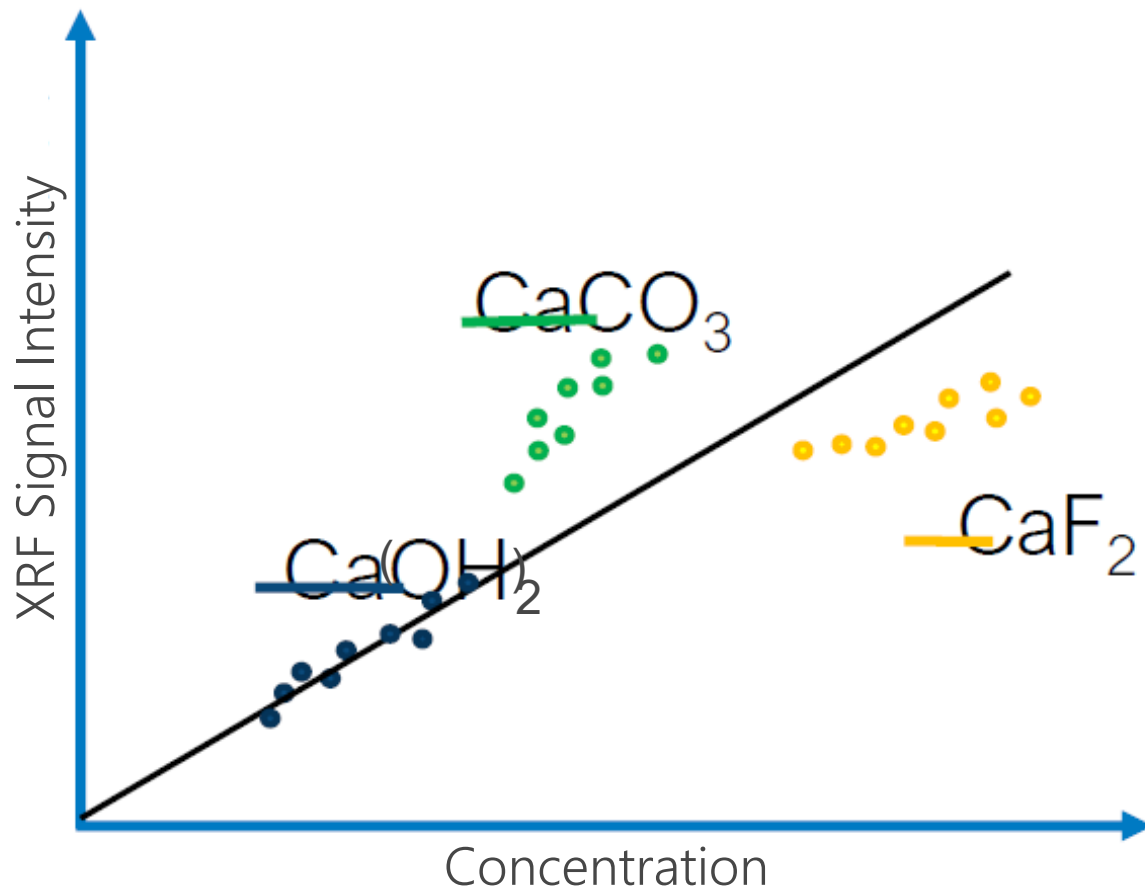




Where are the sources of error and what influence do they have?



Ca calibration in different matrices



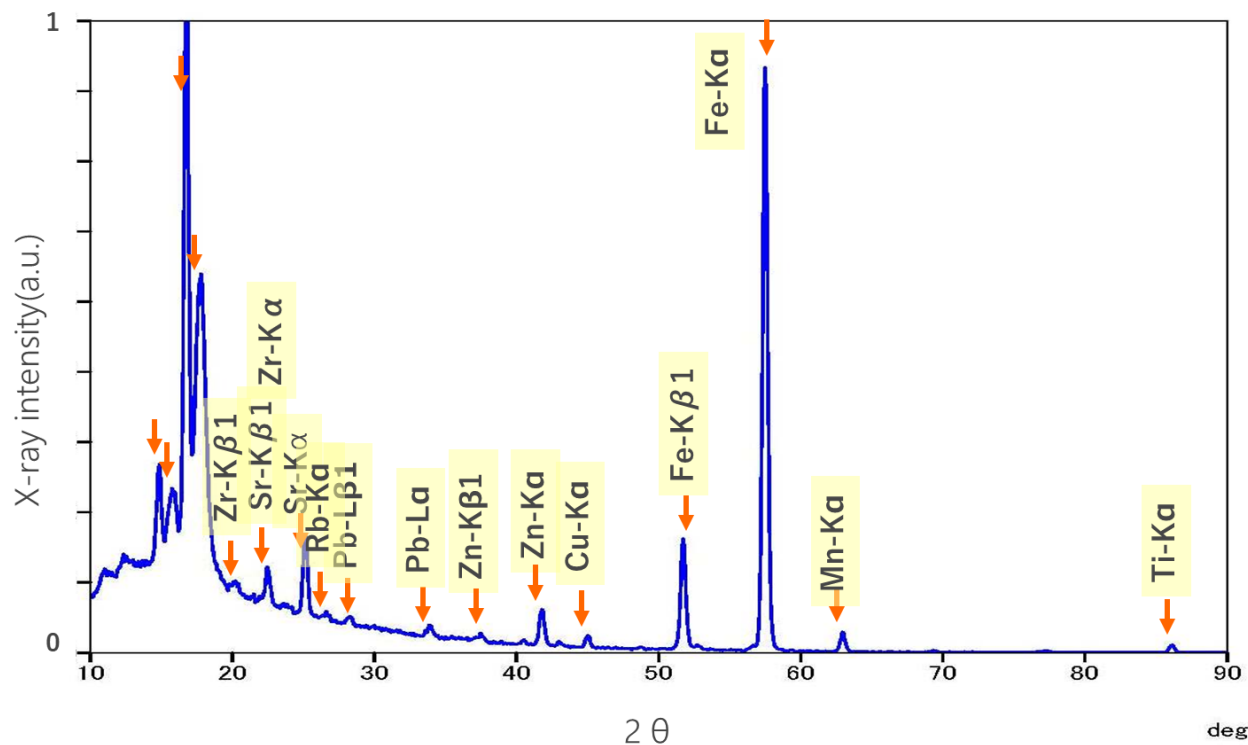
Matrix Customized Calibration

A must for precise analysis results

Strong influence of:

Absorption effects, spectroscopic background and interference

Matrix Compensation Methods on standardization procedures



Fundamental parameter model

- Intensities are compared with a sensitivity library
- Inherent matrix correction for absorption and enhancement effects
- Very good matches possible

e.g. 0.12 % relative deviation for Si in fused beads from geological samples

$$C_{i,u} = \left(\frac{C_{i,r}}{I_{i,r} \left(1 + \sum_j \alpha_{ij} C_{j,r} \right)} \right) \cdot I_{i,u} \cdot \left(1 + \sum_j \alpha_{ij} C_{j,u} \right)$$

Thompson-Scattering

$$E_R = E_P$$

Compton-Scattering

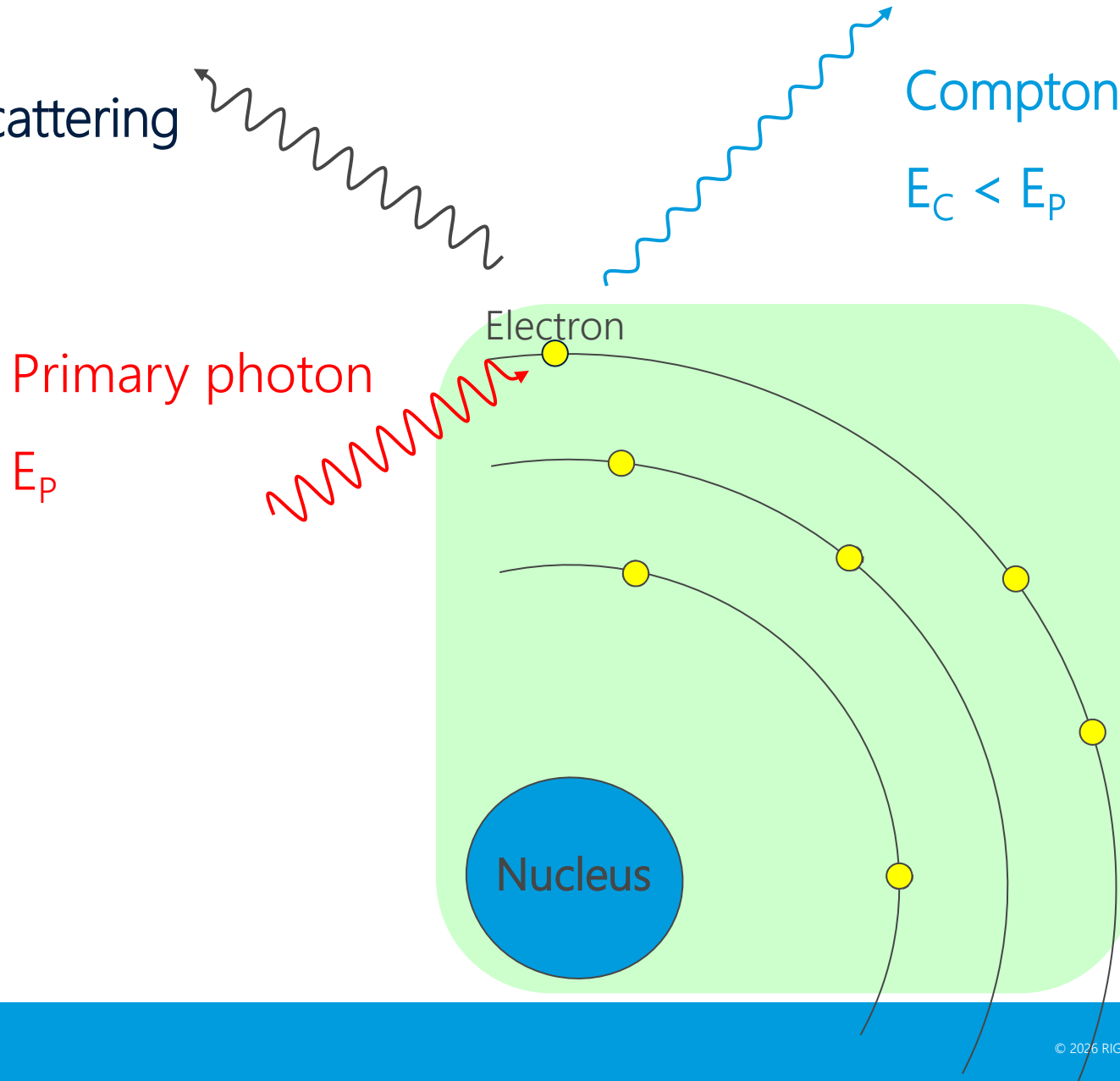
$$E_C < E_P$$

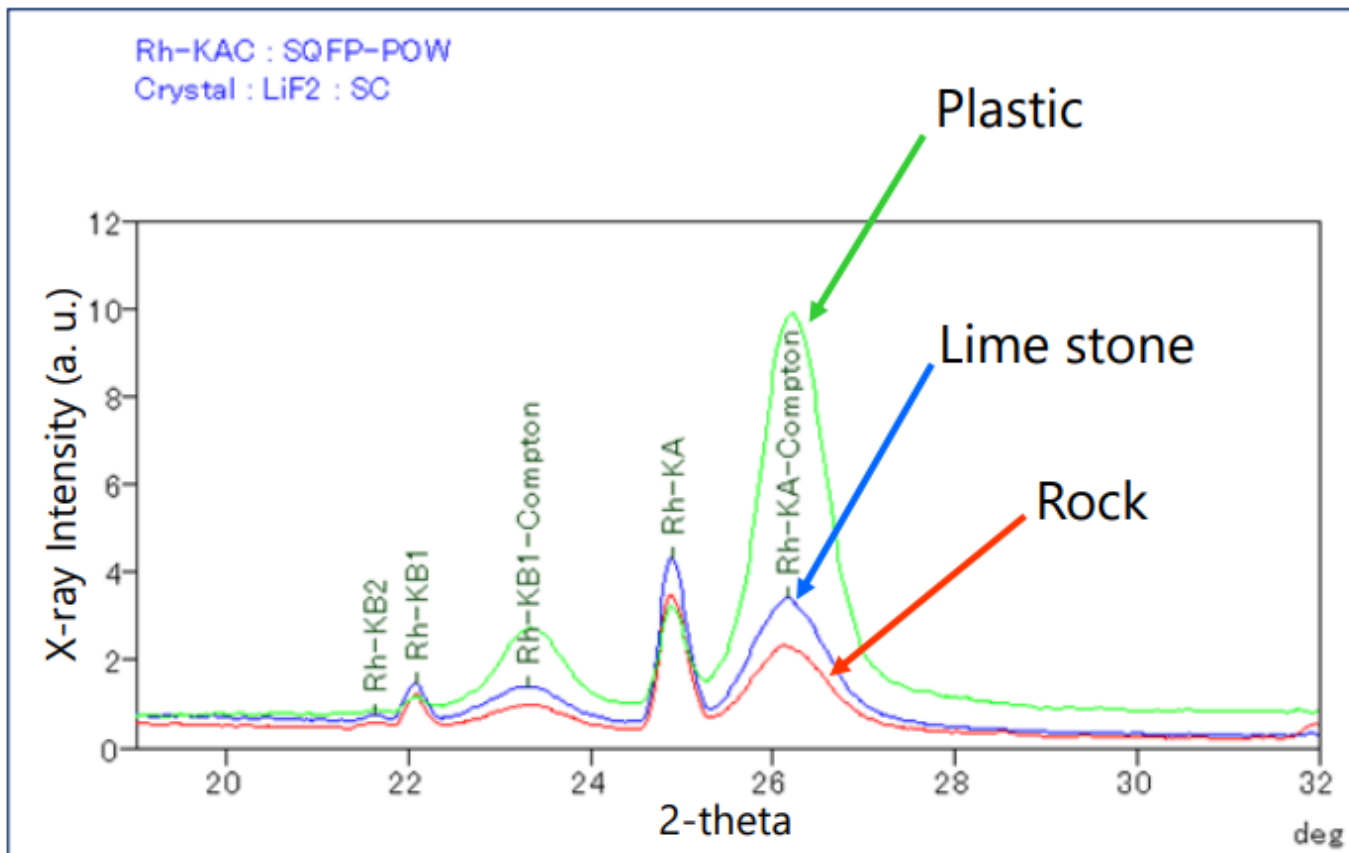
Primary photon

$$E_P$$

Compton Scattering

With a low atomic number and high photon energy, the probability increases





Measurement of Compton Scattering

- Indirect determination of light matrix components such as C, H and N
- Advantage over direct determination:
Measurement using high-energy radiation = large penetration depth
thus more robust measurement
- Enables good approximations using standard-less semi-quantitative methods

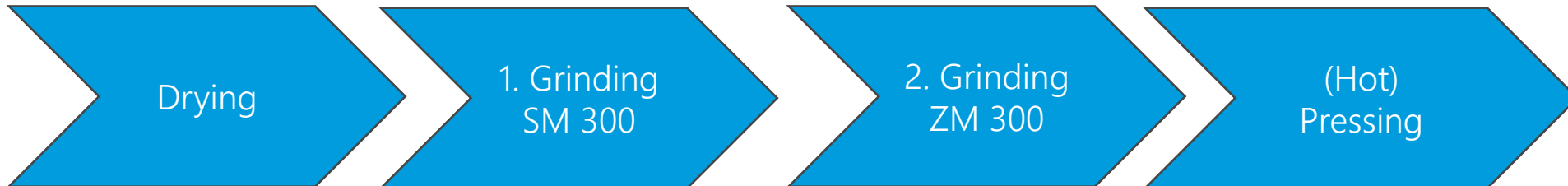


Problems with elemental quantification in plastics

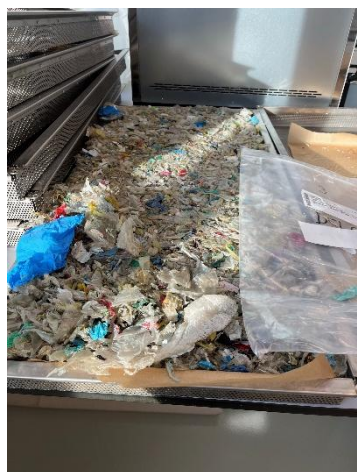
- Digestions with high material use
- Fluctuating matrix makes matrix-matched calibrations difficult
- High workload and high susceptibility to errors

XRF Fundamental Parameter Method

- Minimal sample preparation Measurement of coarse powder pours possible
- Minimal use of materials
- Fast measurement and high flexibility Enables higher sample throughput
- Awareness of the limitations of the method required



Finished sample for XRF analysis



Up to this point, work steps are identical for e.g. ICP

Tyre Derived Fuel (TDF)

Scrap tyres as secondary fuel in cement production

Pre-cut sample



After cryo-grinding



Heterogeneous sample

Multiple determination necessary
→ A and B Sample

Tyres	Sample A	Sample B
	Mass %	Mass %
Na	0.08	0.19
Mg	0.05	
Al	0.13	0.09
Si	0.21	0.21
P	0.03	
Si	0.84	0.89
Cl	0.24	0.18
K	0.09	
Ca	0.22	0.14
Cr	0.09	0.08
Fe	0.17	0.06
Cu	0.08	
Zn	3.56	2.85
FP Balance	94.21	95.31
Total	100	100



Conventional analytical methods:

- Heterogeneous, fluctuating material flows require a high level of analytical effort
- Optimization of sampling and sample preparation

Standard-less semi-quantitative XRF provides

- Flexibility for different matrices
 - Time savings in sample preparation
 - High throughput
- More capacity for better screening

Questions to Plastic recycling?





Application 2

Analysis of organic fibers

Straw as green feedstock in industrial applications

Application:

Straw and other organic fibers are used in various industrial applications as alternative to fossil feedstocks:

→ BtL Processes: Bio to Liquid

Synfuel or platform chemicals (SynGas)



Challenge

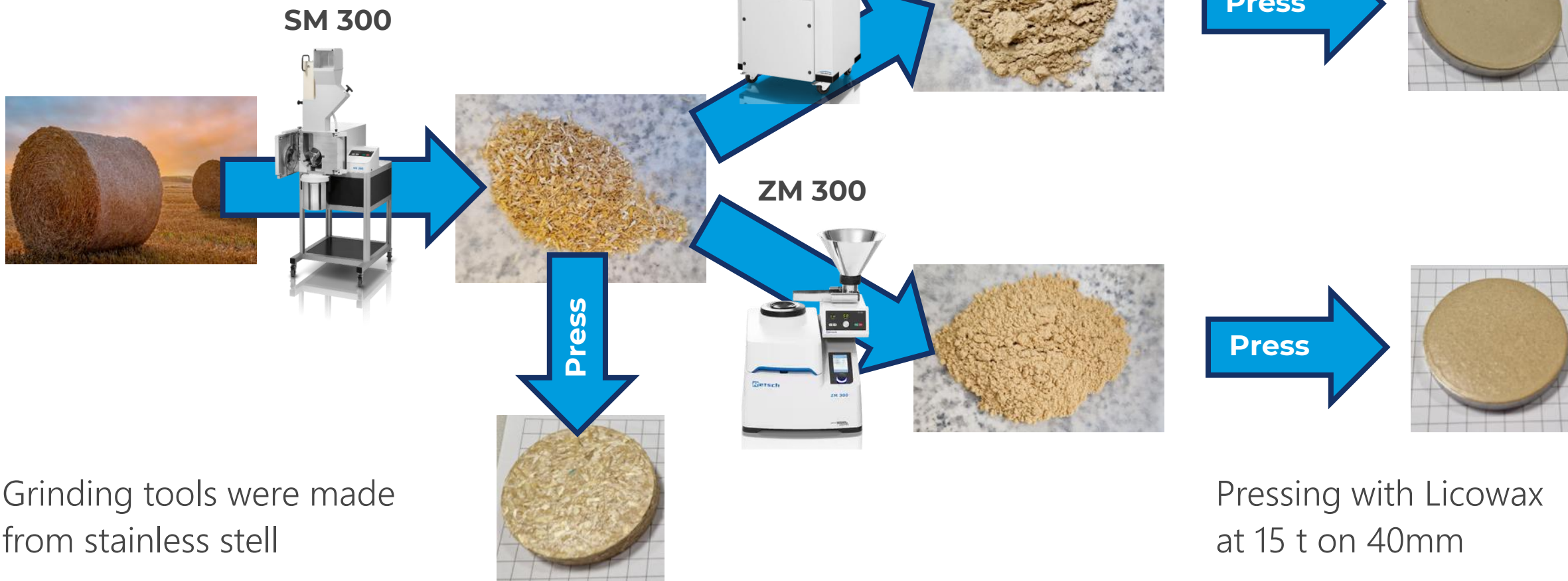
- Humidity adheres well to the fibers
- Agricultural product with strongly varying quality
→ light element concentration influences pyrolysis behaviour
- Multilayer structure
→ Influence on the sample preparation?

XRF Solution

- Easy Analysis from grind biomass
- No combustion, extraction or digestion

Fine grinding: Straw sample – choice of mill

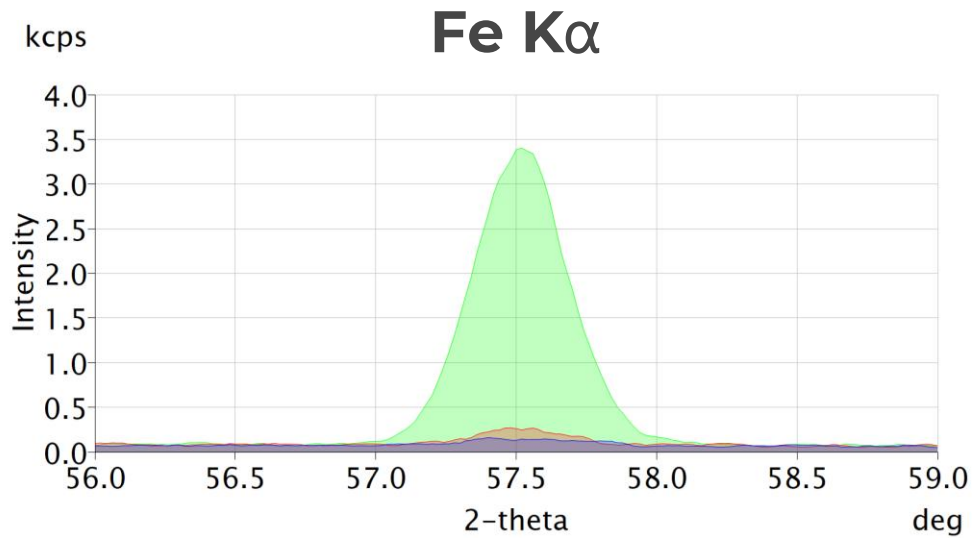
- SM 300 Cutting mill
- ZM 300 Ultra-centrifugal mill
- PM 400 Planetary ball mill



Grinding tools were made from stainless steel

Pressing with Licowax at 15 t on 40mm

Fine grinding: Straw sample – abrasion



Eisenabrieb (Fe)

PM 400 1642 ppm

ZM 300 93 ppm

SM 300 40 ppm

SM 300 Cutting Mill

Coarse crushing, short grinding time, hardly any abrasion

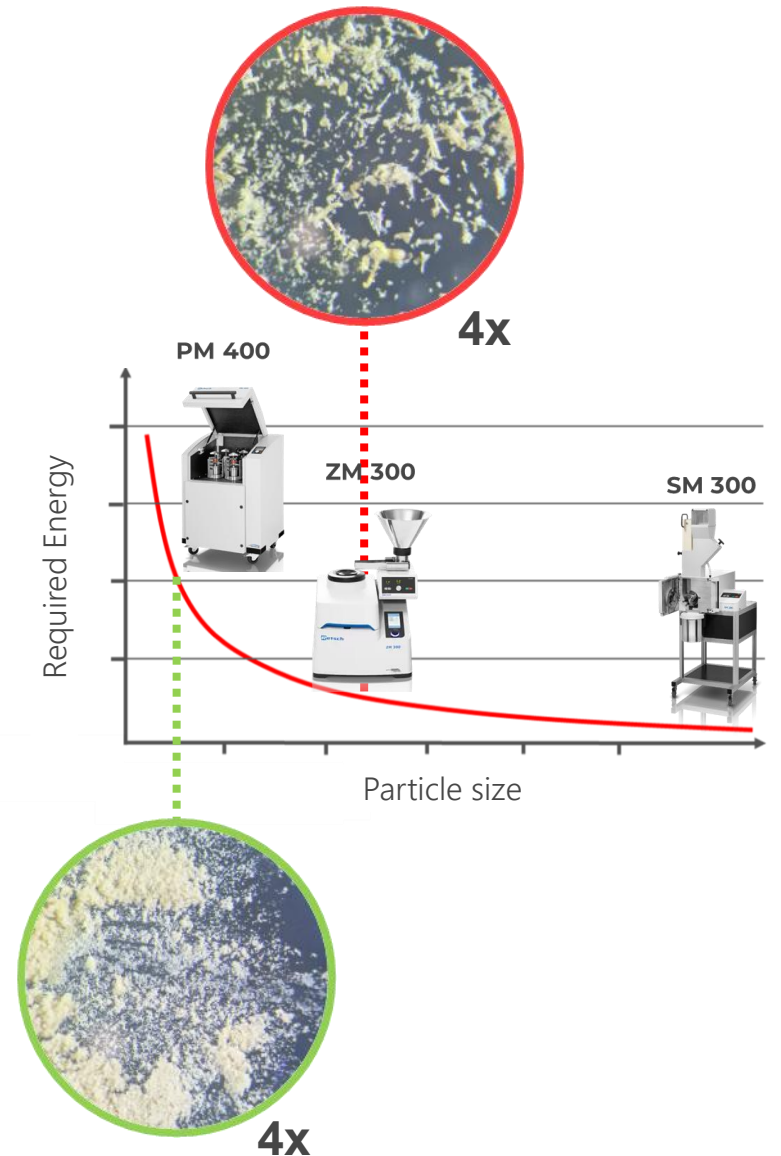
ZM 300 Ultra-Centrifugal Mill

Fast fine crushing, little abrasion, but fibrous powder

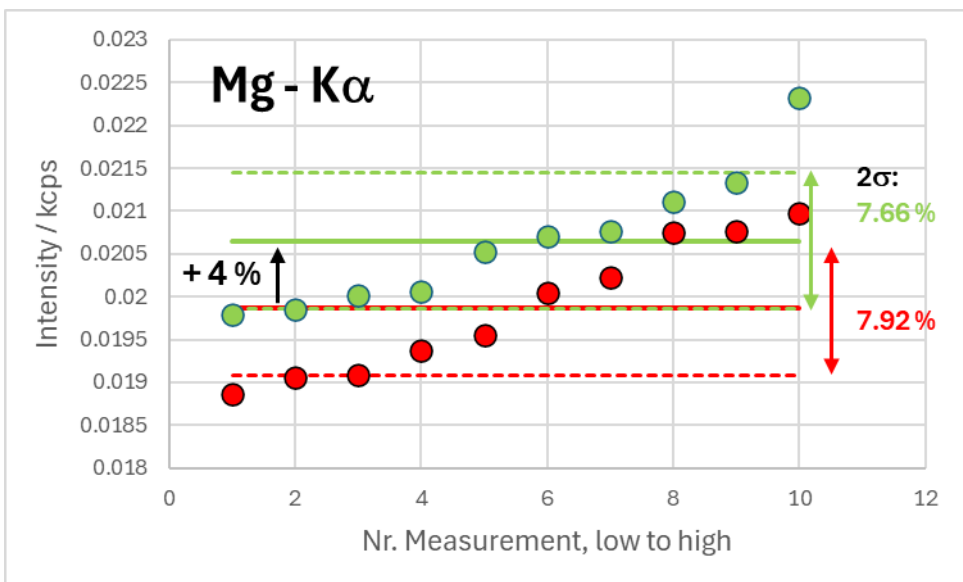
PM 400 Planetary Ball Mill

Long grinding time, with a lot of friction to crush the fibers

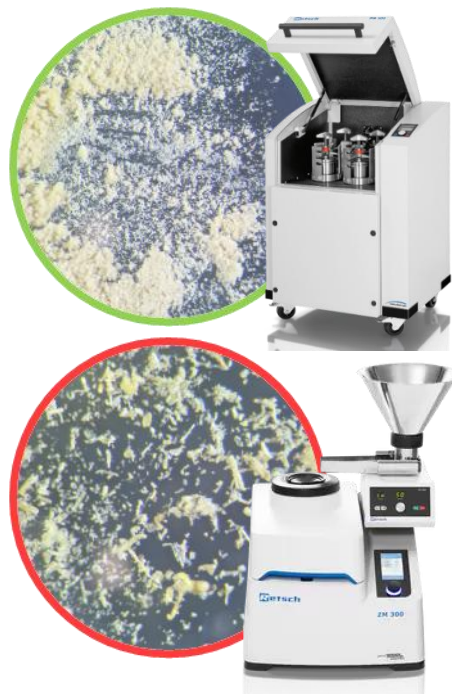
High abrasion, fine powder



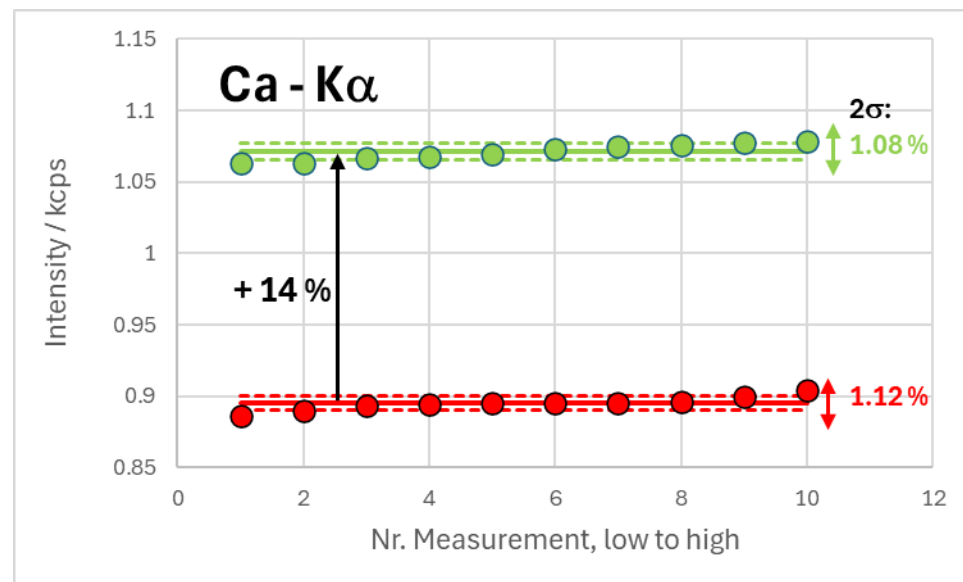
Measurement: 60 s peak
2 x 30 s underground



PM 400



ZM 300

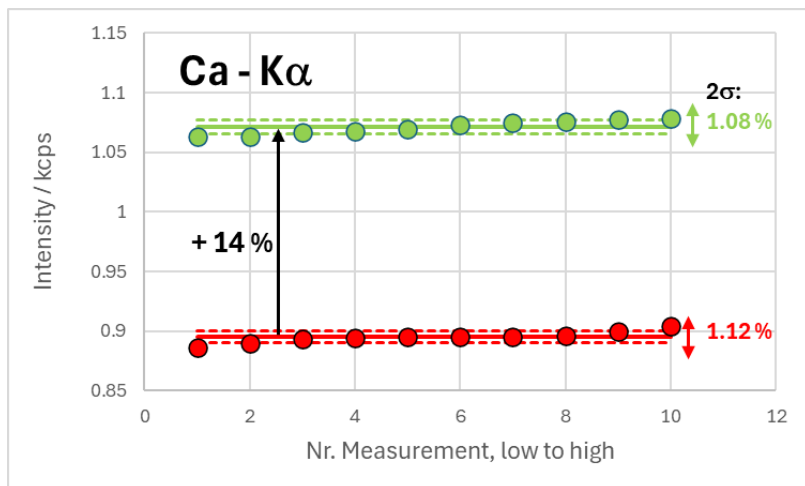


Fine grinding in the **PM 400** planetary ball mill leads to better homogeneity of the sample.

Organic fibers are highly structured materials with anisotropic element distribution

Grinding mitigates this

→ Representative signal



Straight forward analysis with XRF?

- Initial statement: Easy Analysis from grind biomass

Pitfalls need to be identified and avoided

- Anisotropic bio fibers when ground can smear and not reduce particle size
- High power milling equipment required

➔ More about the importance of sample preparation after the lunch break



Questions to Analysing organic fibers?



Application 3

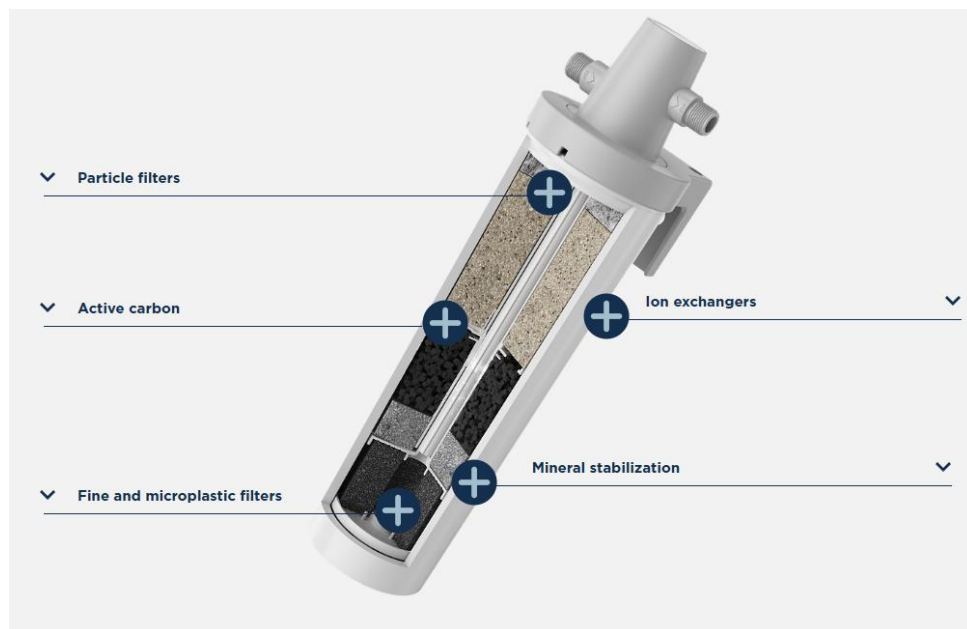
Calibrating loose & wet powder

On the example of ion exchange resin

Application:

Water purification systems are prone to bacterial infestation

Ion exchange resins which release antimicrobial Ag⁺ ions are used to inhibit bacterial growth



Challenge

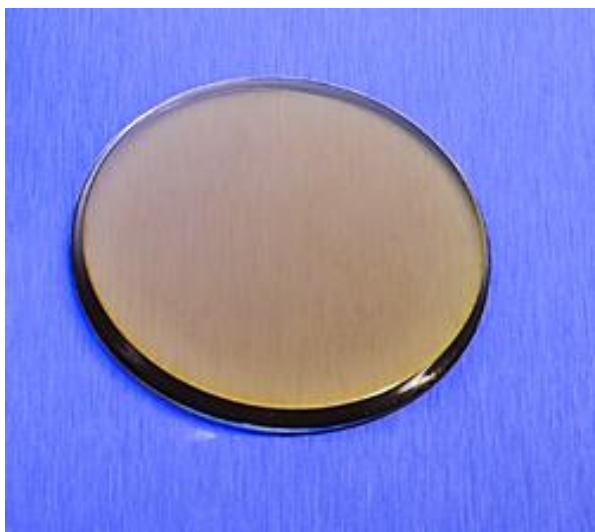
- Sample is strongly hygroscopic
- Digestion attempts by the customer failed
- Resin particles are too big to press and too ductile to easily grind down

XRF Solution

- Calibrate for the wet loose powder in a cup

How should the ideal sample look like:

- Homogeneous
- Dry
- Smooth surface / fine particle size
- No air gaps
- Infinitely thick



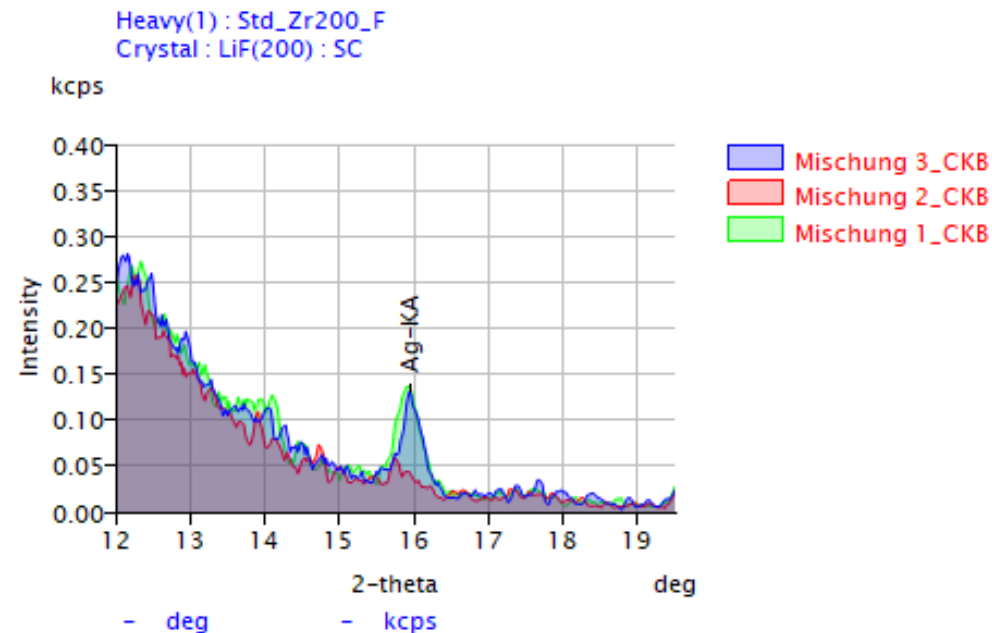
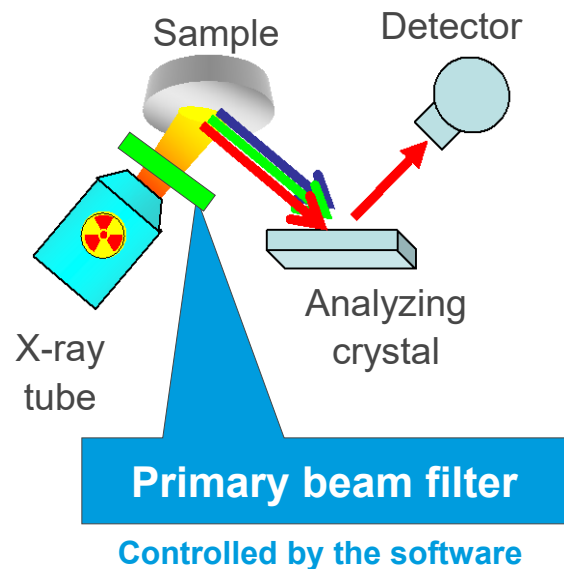
How does the sample look like?

- Wet (hard to dry)
- Wet powders shouldn't be pressed

→ Loose powder calibration required



Analysis of Ag⁺ in Ion Exchange Resin



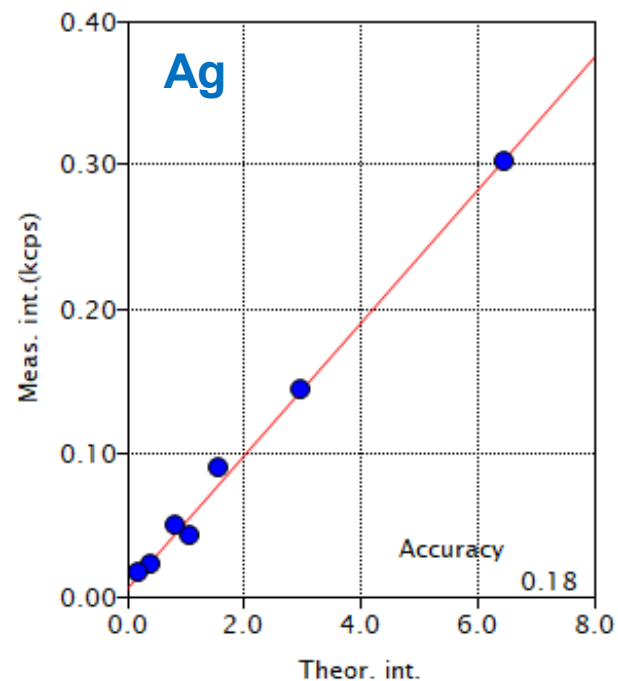
Ag has a direct line overlap with the characteristic lines from the Pd tube
 Zr filter (standard) eliminates interfering Pd-K lines and Compton scatter lines from the X-ray tube target. Used for analysis of Cd, Ru, Rh, Pd, Ag, or In.

Good spectral resolution & low background



Regression r^2 0.9974
 Slope 2.168e1
 Intercept -1.233e-1

El. line Ag-KA



● Certified ● Uncertified ● Deselected ◇ Corrected

Ag-Calibration of loose powders.

- Calibration standards from readily available materials

→ Mixture from fresh resin (100 % Ag) and blank resin (0 % Ag)

- FP Calibration mitigates influence of varying matrix scattering

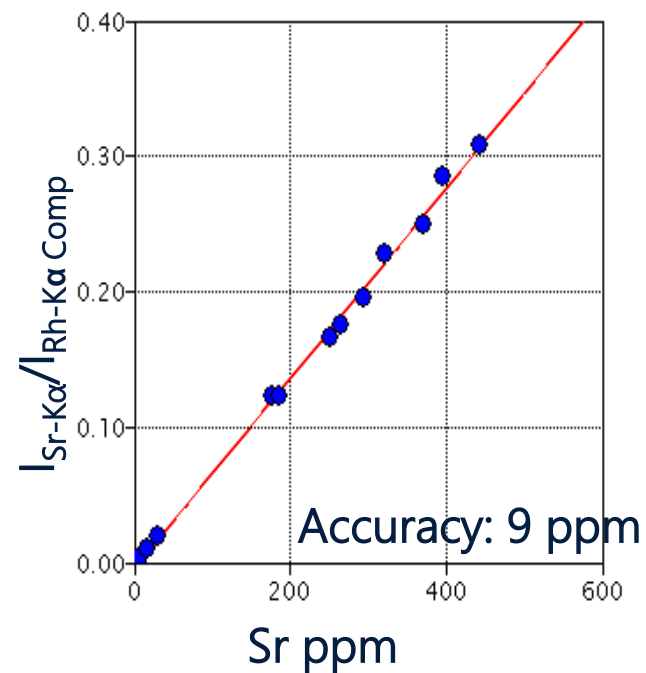
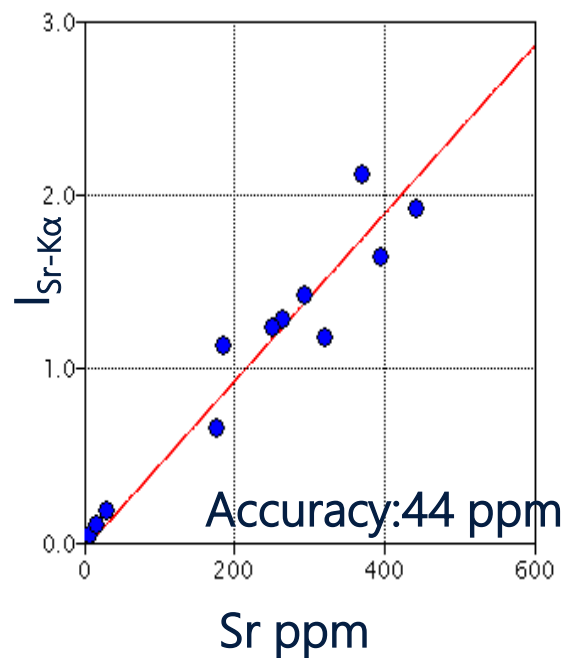
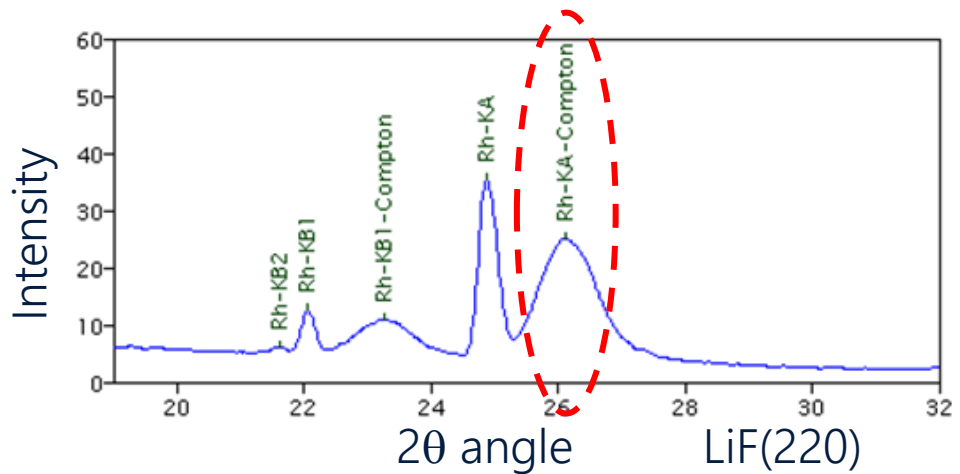
→ Varying water content & particle distribution

Analysis of Sr in rocks

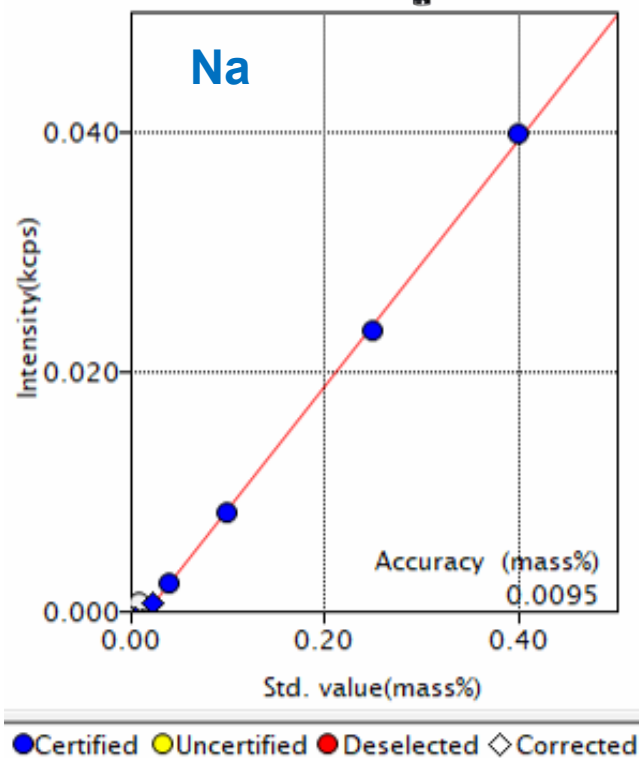
Calibration equation

$$W_i = CI_R + D$$

$$I_R = \frac{I_i}{I_{Rh - K\alpha Compton}}$$



Supermini200

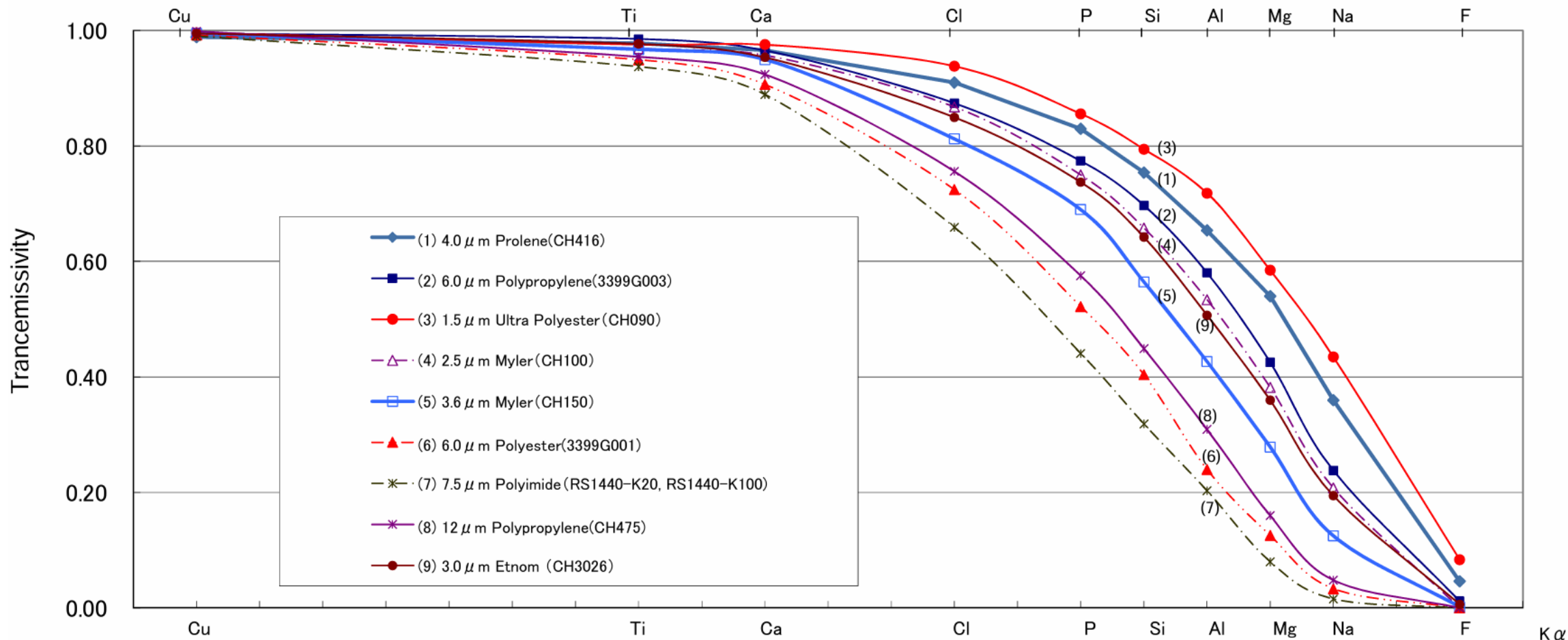


Na-Calibration of loose powders

Low level calibration despite:

- Not ideal sample form
- Measurement in He atmosphere
- Measurement in a cup with a film
→ absorbs >70 % Na signal

Trancemissivity by Experiment





- High quality calibration can be achieved even under adverse prerequisites

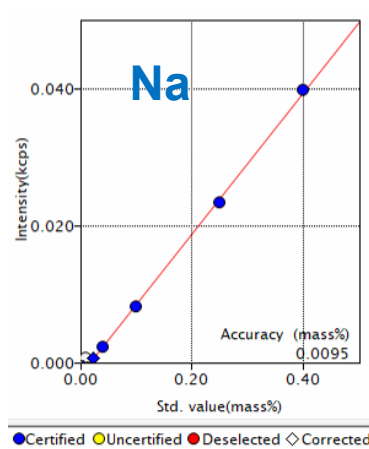
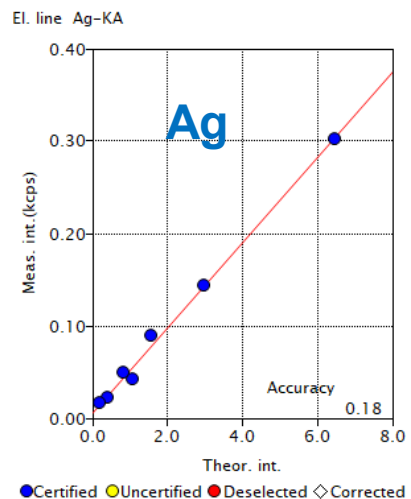
→ Matrix matching is key

- Certain matrix effects can be compensated

→ ZSX Guidance software with FP package

- Benchtop size, full WD performance

→ Supermini200 for low level Na calibration



Questions to Calibrating loose & wet powder



Application 4

Micro Carry for analysis of aqueous solutions

On the example of kaolin as plant
protection agent



Application:

Kaolin suspensions are used in agricultural research as inexpensive biocompatible plant protection

agents:

- Insect repellent (*Drosophila suzukii*)
Used until shortly before harvest
- Reflecting agent to reduce water evaporation in the summer

Application via regular water irrigation systems

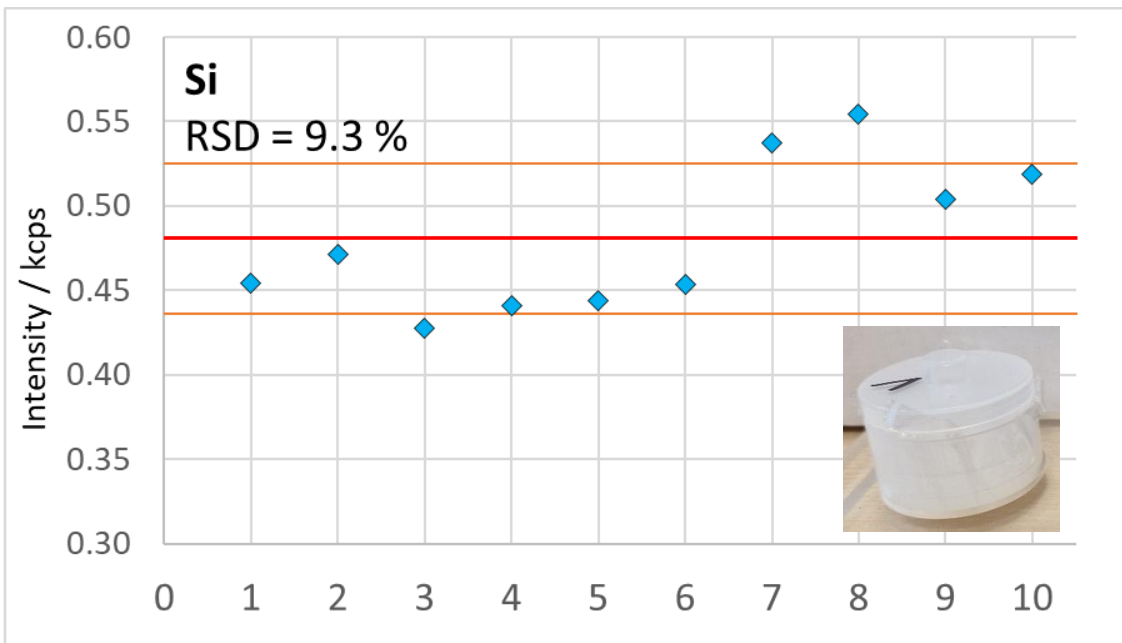
Challenge

- Water spraying tanks need to be checked before regular irrigation
- Kaolin $AlSiO(OH)$ is difficult to digest for ICP analysis
- Suspensions are not stable



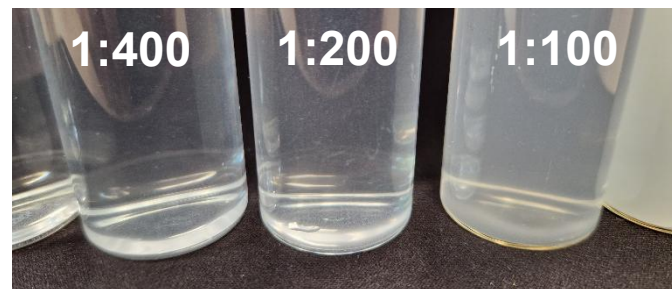
XRF Solution

- Measurement via preconcentration on filter materials (Micro Carry / Ultra Carry)



before homogenizing

after homogenizing

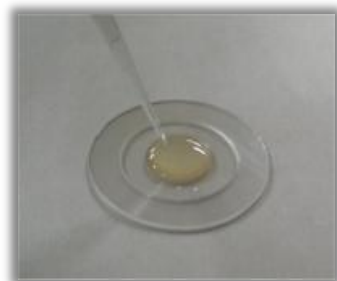


Kaolin suspensions:

- Sedimentation is an issue for this sample
→ Enrichment in the analytical layer
- Signal intensity dependend on the time between preparation and measurement
 - Each cup prepared on the left was measured twice (see graph)

- Poor repeatability

→ Not fit for calibration!



Take a representative sample

Pipette 10 to 500 μL on the filter

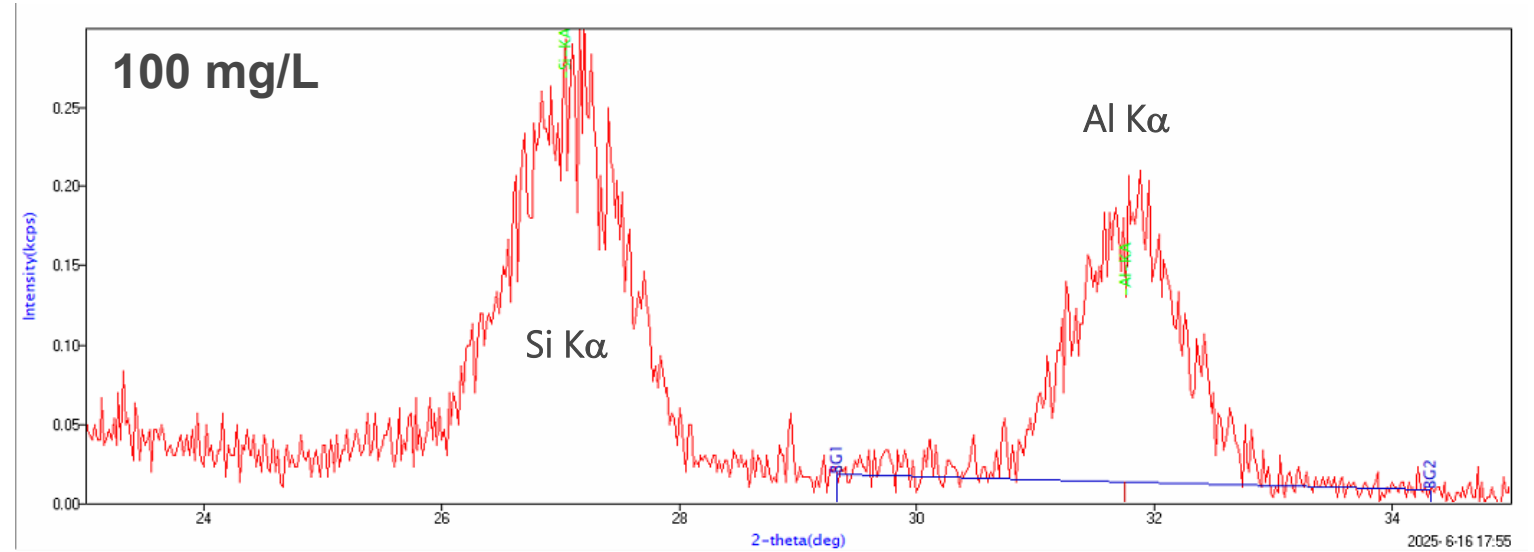
Dry at 45° C ideally under vacuum

Place in sample holder and analyze

Supermini200



- MicroCarry ($\leq 200 \mu\text{L}$) and UltraCarry ($\leq 500 \mu\text{L}$) allow precise analysis of small sample volumes
- High precision with low limits of detection
- Undesirable matrix is removed by drying
- Left over is a near perfect sample thin homogeneous layer on the surface of the filter
- No film needed \rightarrow high sensitivity of light elements
- ICP standards for calibration



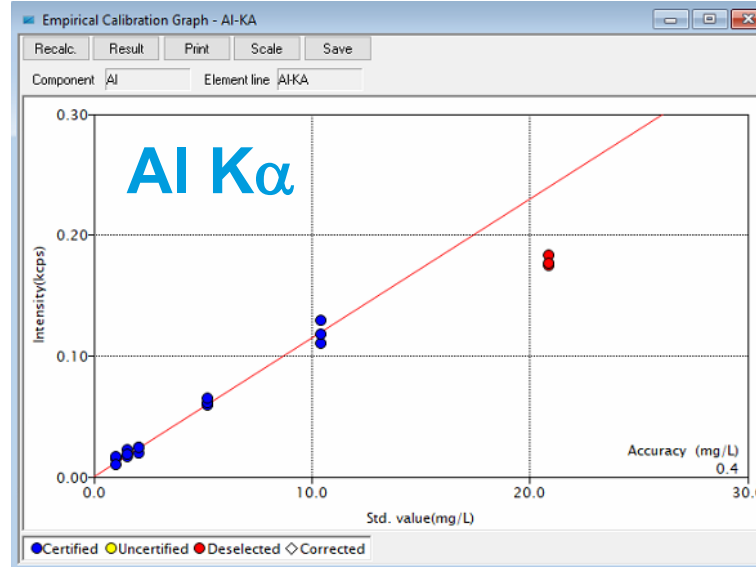
Calibration method:

- Empirical calibration on stock solutions of defined kaolin content
- 3 samples per standard
→ minimal spread

→ Here: accuracy for Kaolin: 1.9 – 2.1 mg/L

→ LOD (Kaolin) = 0.6 – 0.7 mg/L

Further standards and longer measuring times can further improve LOD and accuracy



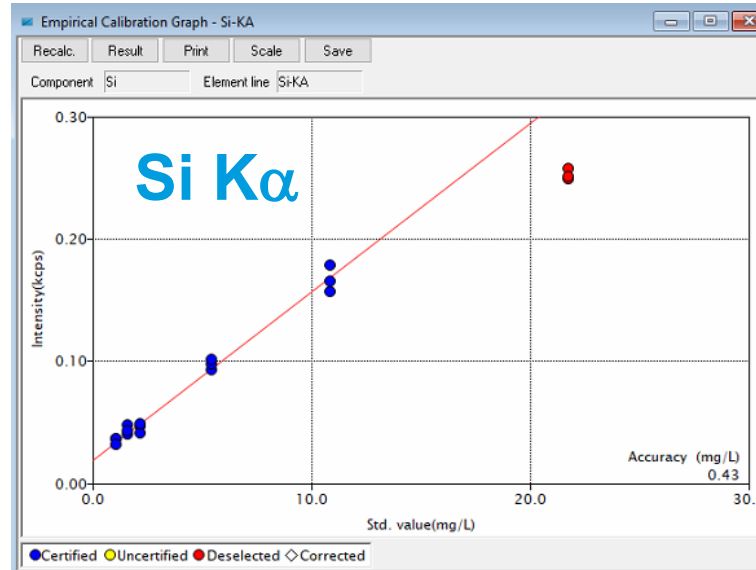
Al

Standard 6 not considered

Corr. Factor 0.994

Accuracy ± 0.40 mg/L

LOD (Al) ~0.13 mg/L



Si

Standard 6 not considered

Corr. Factor 0.994

Accuracy ± 0.43 mg/L

LOD (Si) ~0.17 mg/L

For the dilution of 1:400 the Al concentration converts to

$$c(\text{Kaolin}) = 25.83 \pm 0.29 \text{ mg/L}$$

vs. the standard value given at 25 mg/L

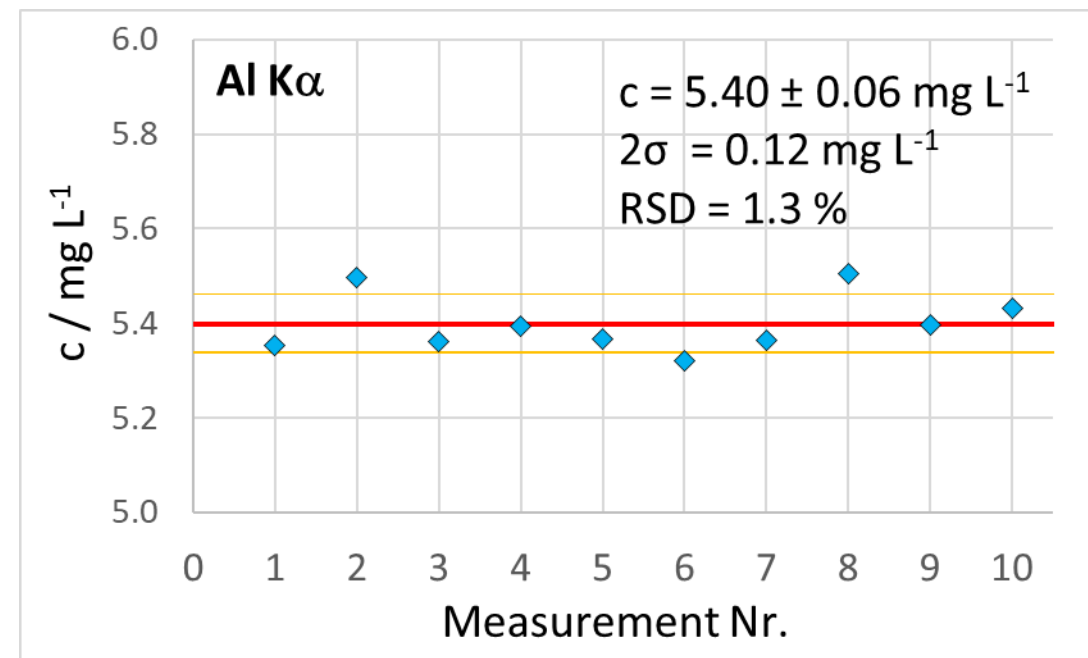
The conversion of Al Ka and Si Ka signal intensity to Kaolin concentration can be done automatically by the software (as two independent values)

→ High accuracy provided by the instrument

→ High precision

can be further improved with more standards,
improved preparation, maybe certification

Standard	4(a)
Dilution	1:400
c(Kaolin)	25 mg/L
c(Al)	5.23 mg/L



Summary: MicroCarry

- Excellent for low limits of detection in samples with volatile matrix
- Low sample volumes for scarce sample materials
e.g. Enzyme quantification in life science
- No need for digestions
- Easy Calibration & cheap consumables

Questions to Analysing with the Micro Carry filter?

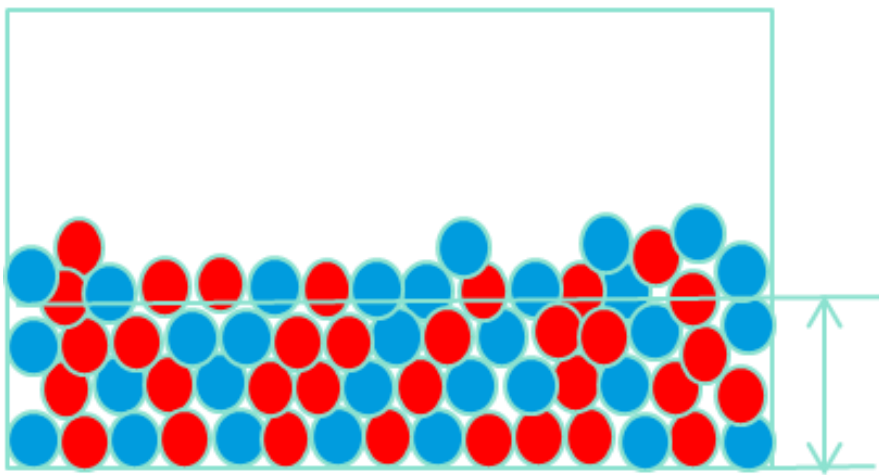
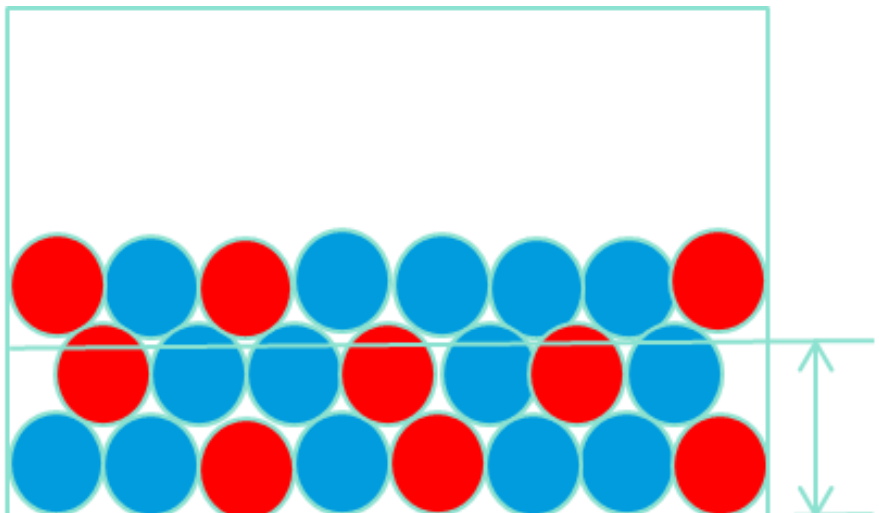


Influence of particle size on the XRF result

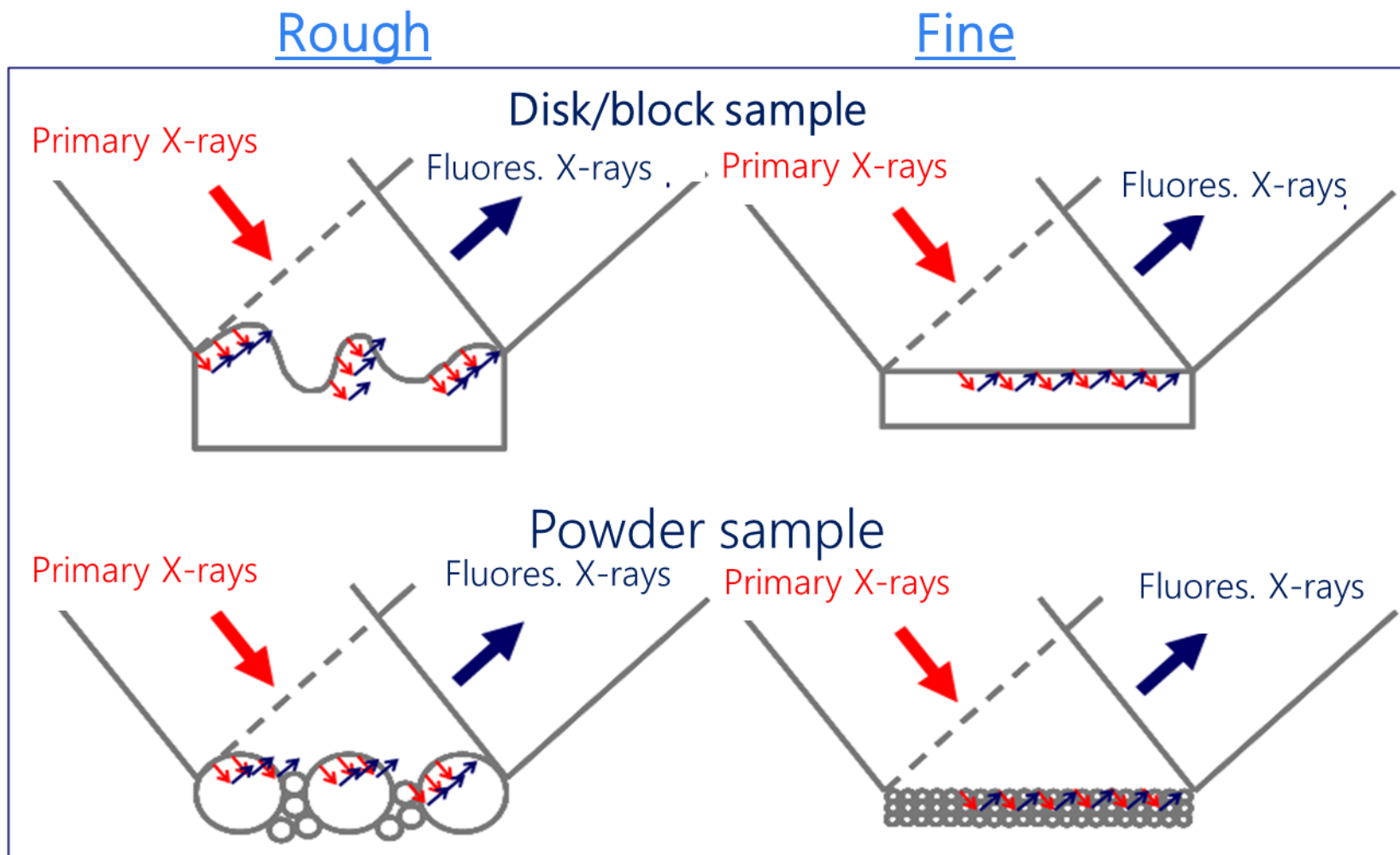
On the example of lime stone



Why is sample preparation / grind important?



- The required particle size is directly correlated with the thickness of the analytical layer
 - If the particles are too coarse, the analyzed layer may no longer be representative
 - Reduction of particle size leads to a denser, more homogeneous and more representative layer
-
- XRF is a relative method, the intensity must be correlated with the concentration
- Matrix-matched calibration is required



Disadvantages of rough surfaces

- Absorption effects, especially with light elements
- Irregular Scattering
- Orientation effects
- Especially disadvantageous when the surface finish fluctuates

Loose Powder



Qualitative and
Semi-quantitative
applications

Pressed Powder

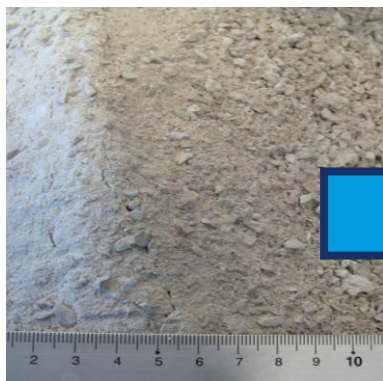


Quantitative Process Control
Matrix matched calibrations

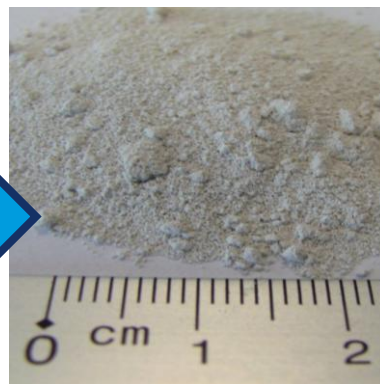
Fused Beads



Reference method
Matrix independent
calibrations



Beispieldarstellung

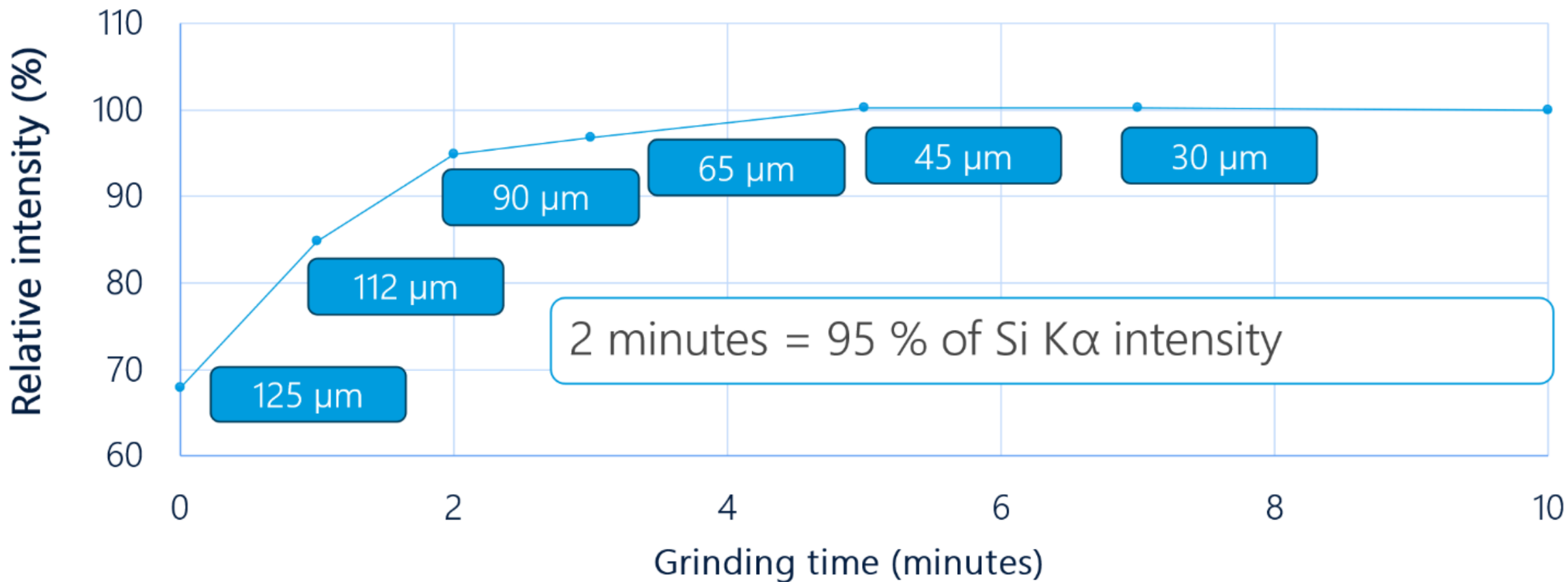


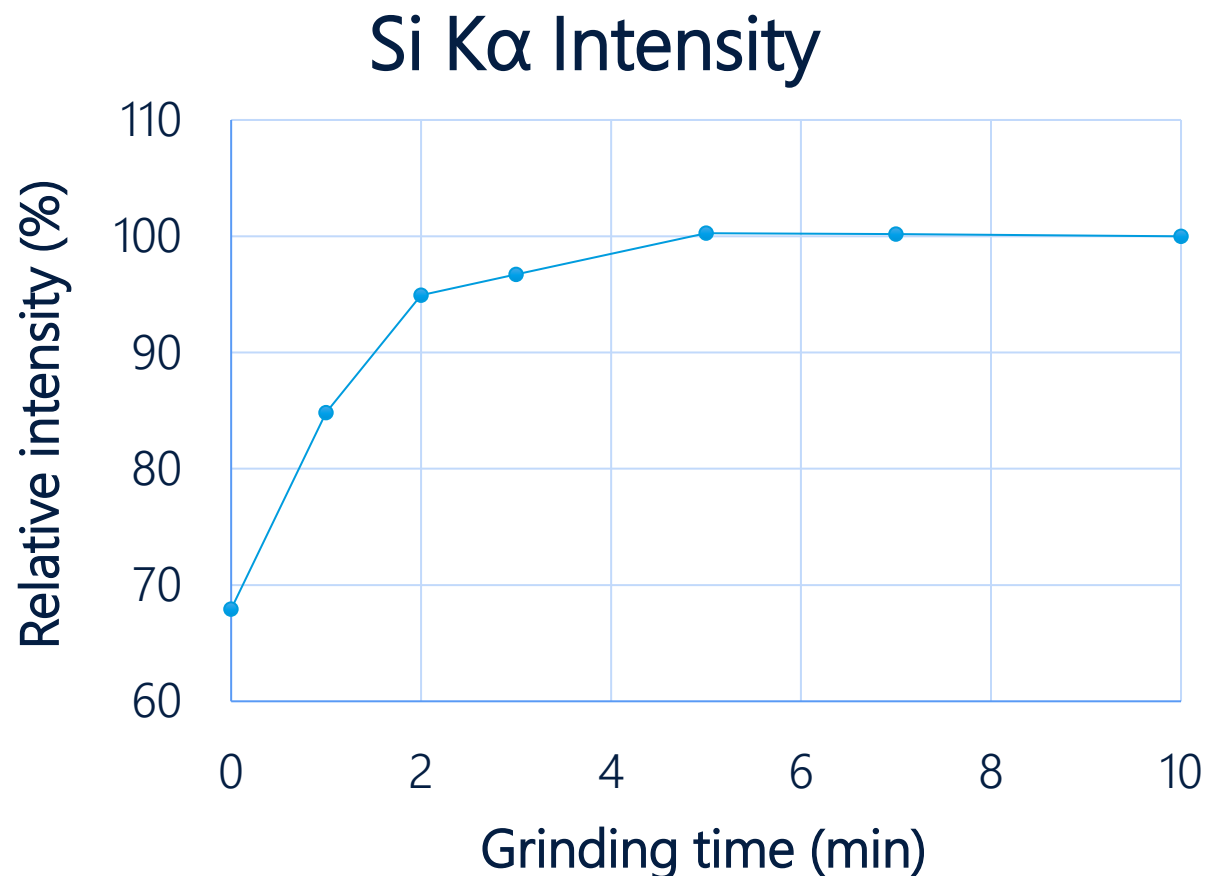
Beispieldarstellung



- Sample limestone powder
- Element Si
- Line $K\alpha$
- Disc Vibrating Mill with Tungsten Carbide Vessel
- Grinding time 1 to 10 minutes
- Constant pelleting parameters
- Constant XRF Parameters

Si K α intensity based on particle size (grinding time)



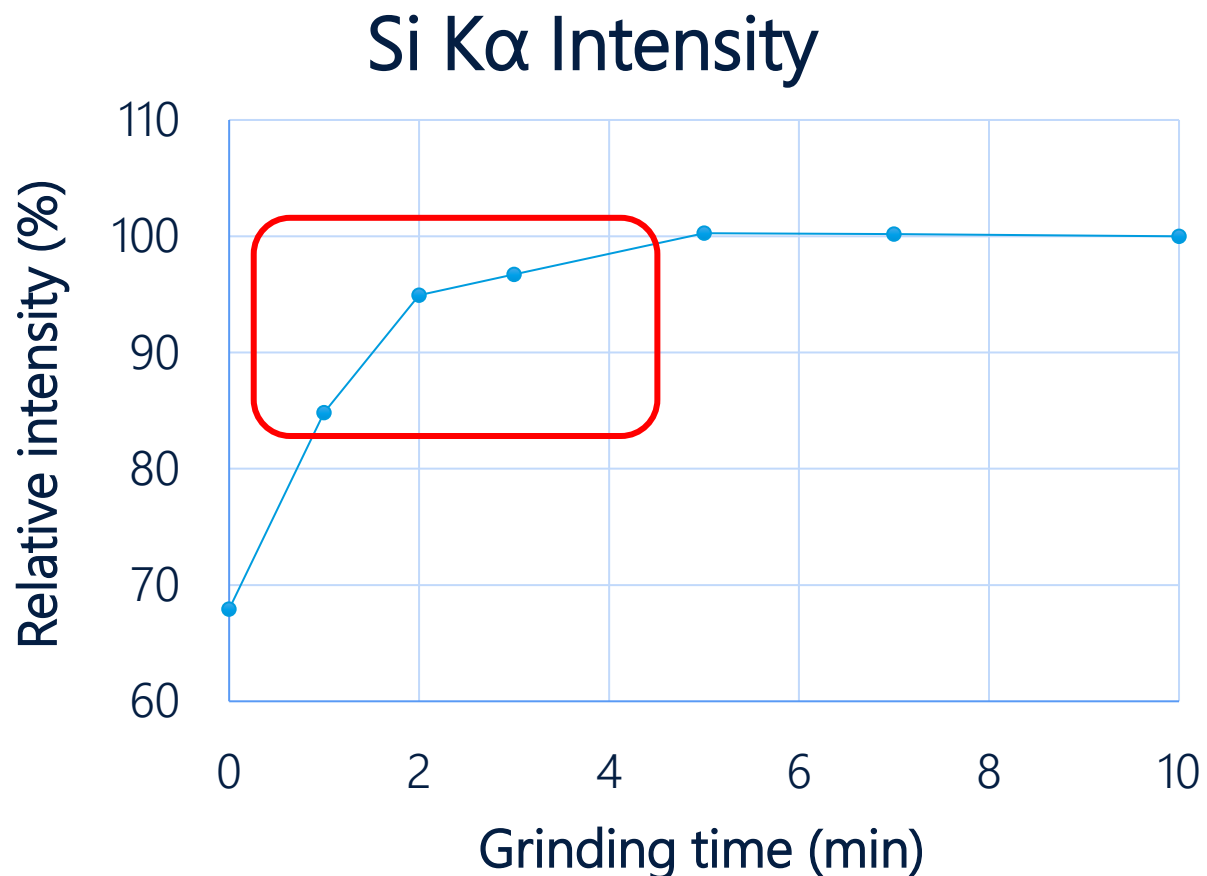


Observation

From 0 min to 10 min, the intensity increases by about half

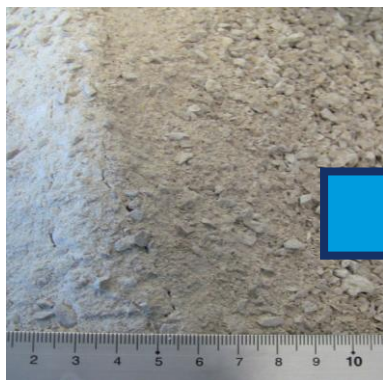
The plateau (100 % relative int.) is reached after 5 minutes

Grinding too short can lead to a loss of intensity, after 1 min only 85 % relative Int. is reached

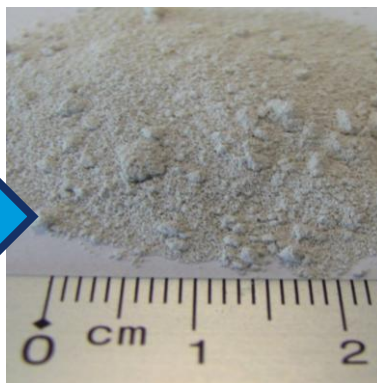


Recommendation

- Grind as long as necessary and as short as possible here 5 min
- Constant / Reproducible grinding time
→ Routine samples and calibration standards
- New sample preparation?
→ new calibration
- Use of programmable mills increases reproducibility



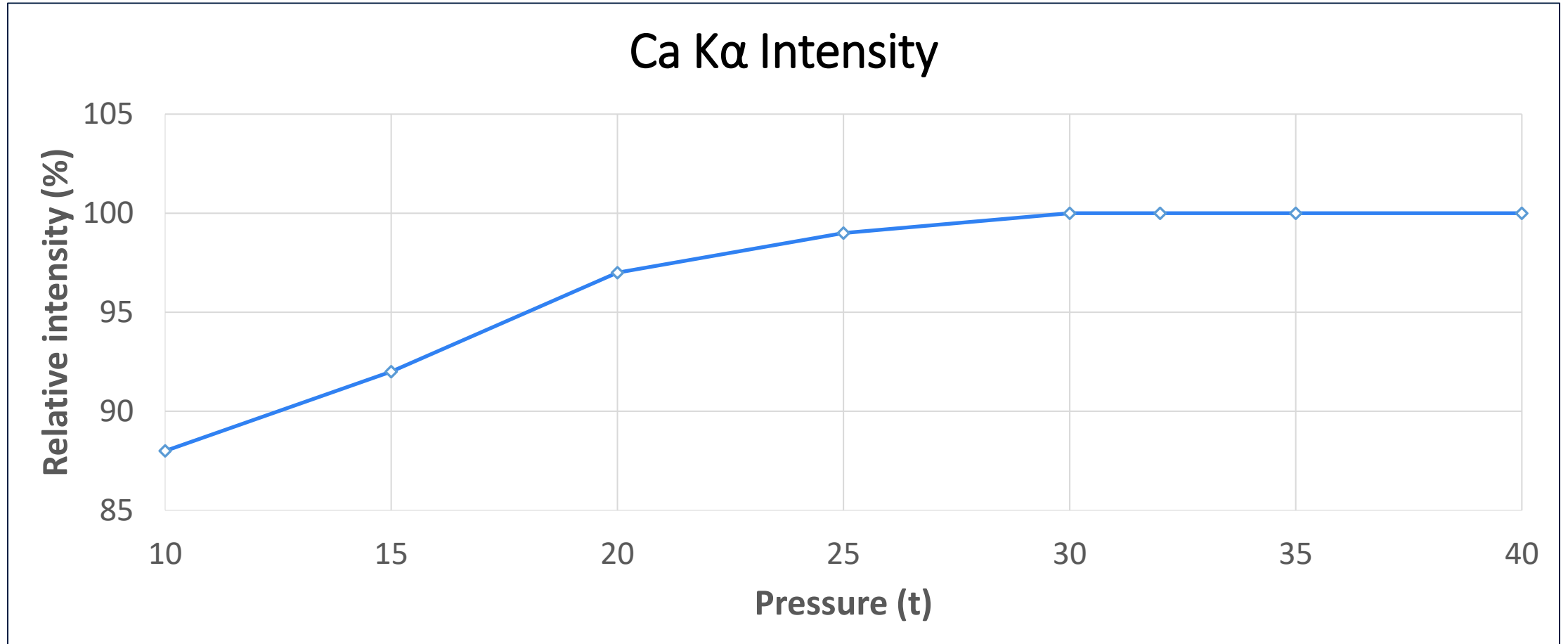
Beispieldarstellung

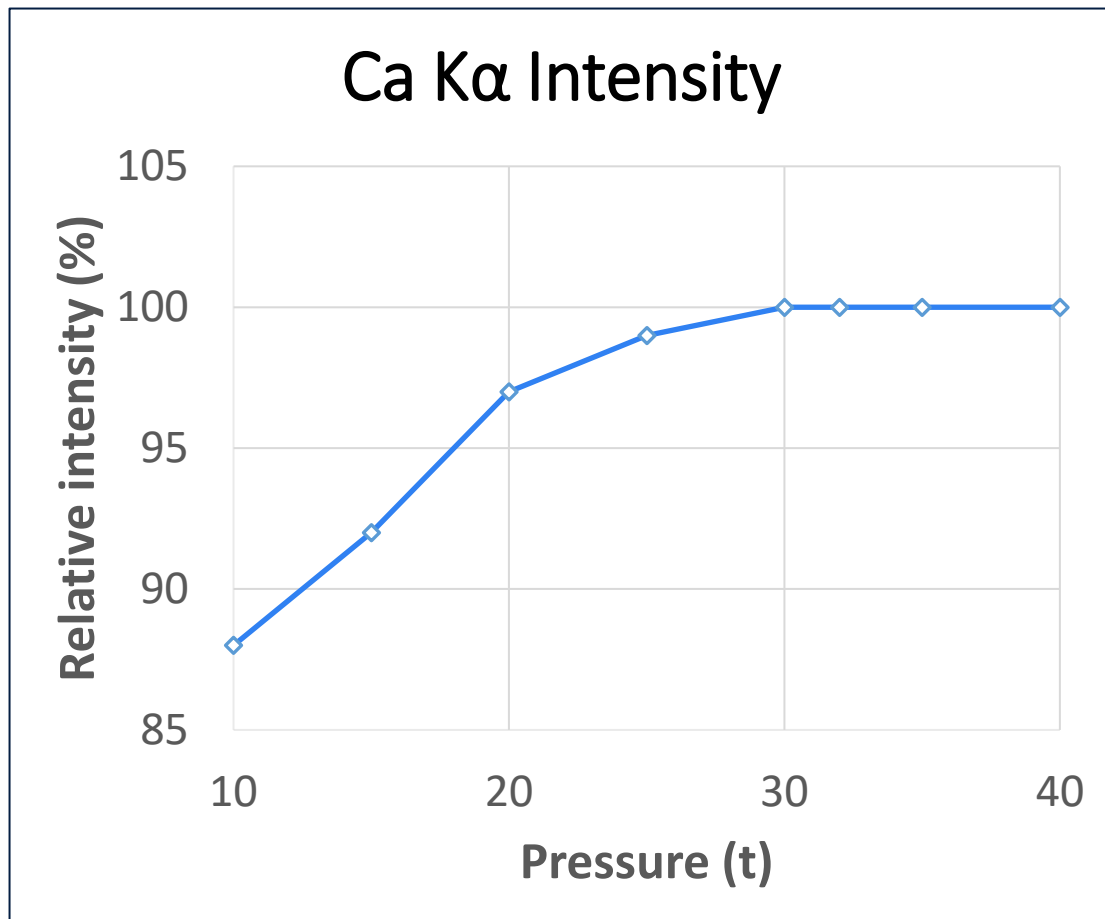


Beispieldarstellung



- Sample limestone powder
- Element approx
Line $K\alpha$
- Disc Vibrating Disk Mill with Tungsten Carbide Vessel
- Press pressure 10 – 40 t
- Constant grinding parameters
- Constant XRF Parameters



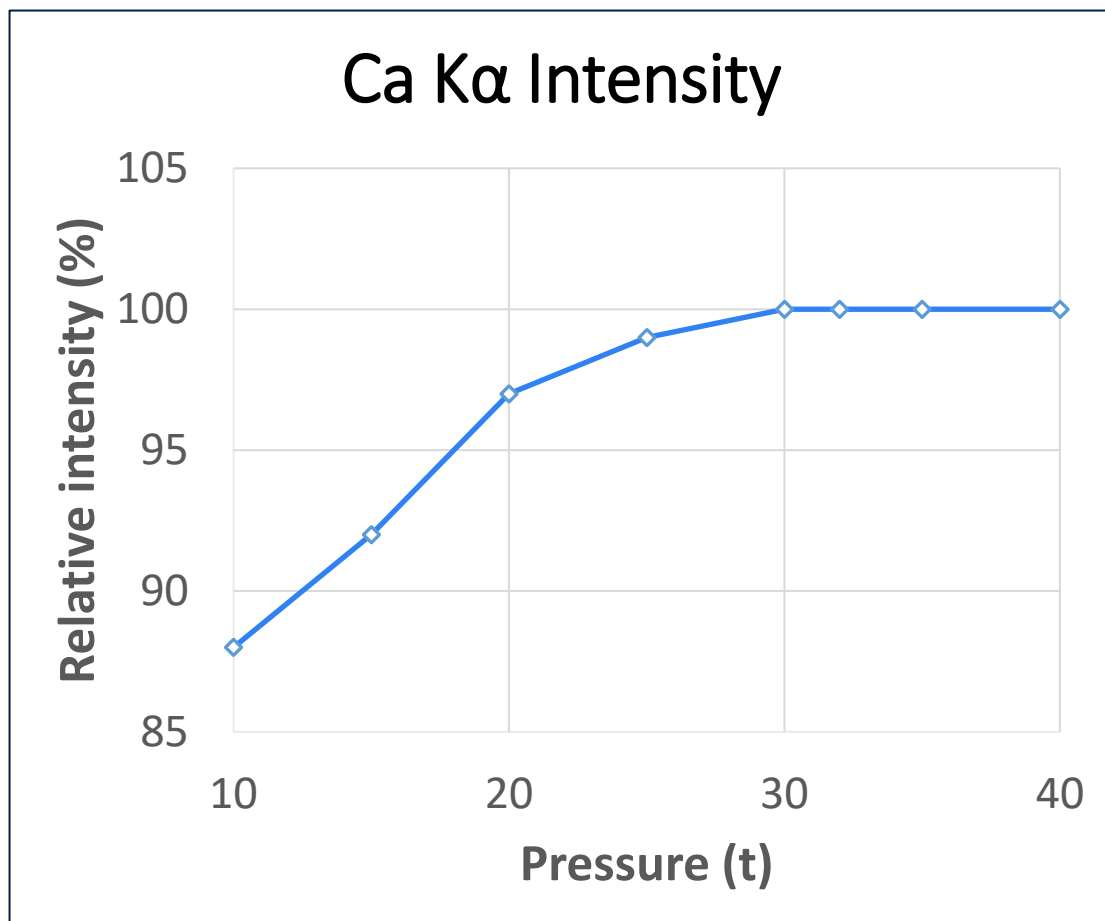


Observation

From 10 t to 40 t, the intensity increases by 12 % points

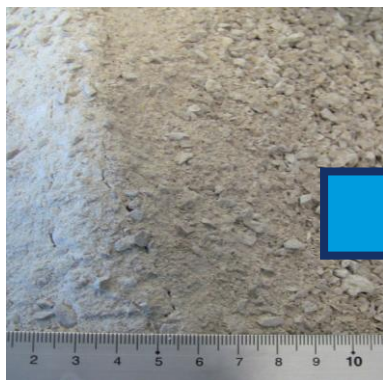
The plateau (100 % relative int.) is reached at 30 t

Pressing at too low a pressure can lead to a loss of intensity

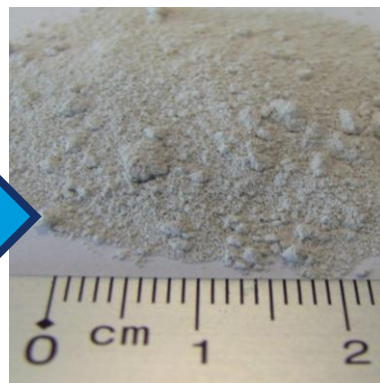


Recommendation for action

- Urgent consideration of pelleting parameters in method development
- When only a smaller press is available, reproducibility is even more important
- Use of programmable presses increases reproducibility



Beispieldarstellung

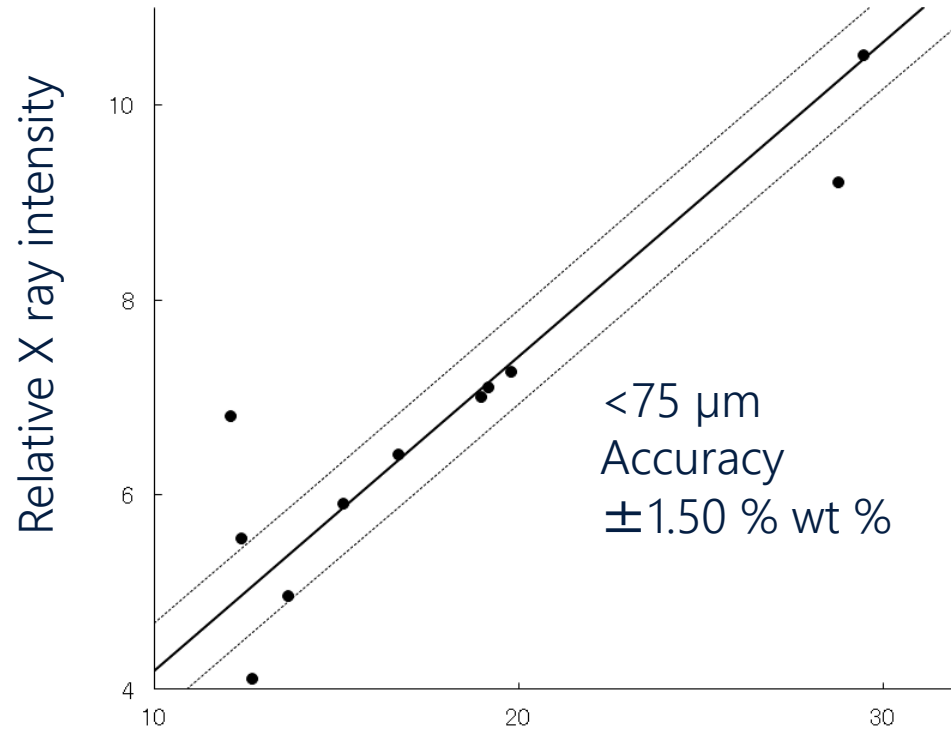


Beispieldarstellung

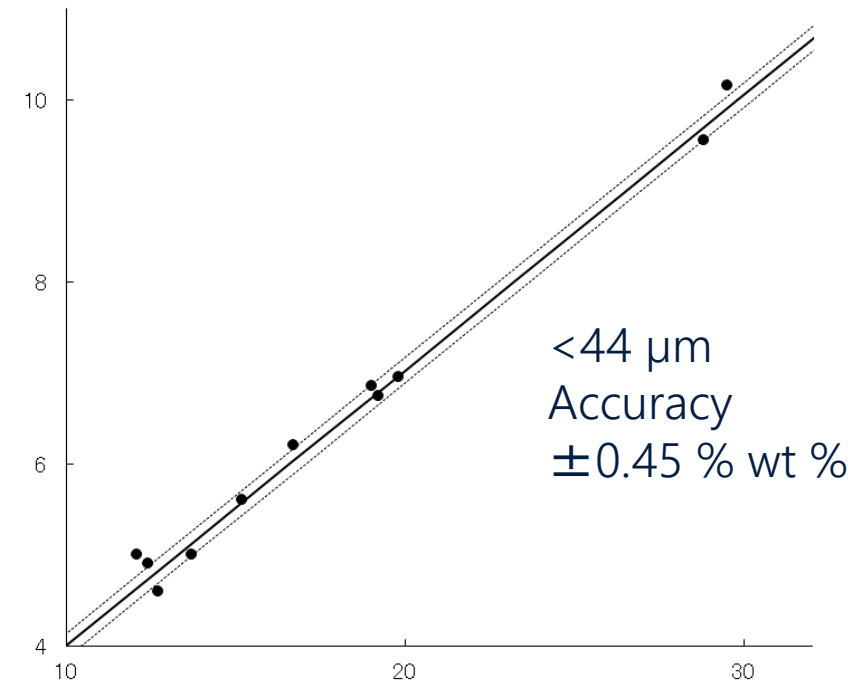


- Sample clay
- Element Al as Al₂O₃
- Line K α
- Disc vibrating mill with
 - Standard Set A < 75 μ m
 - Standard Set B < 44 μ m
- Constant pelleting parameters
- Constant XRF Parameters

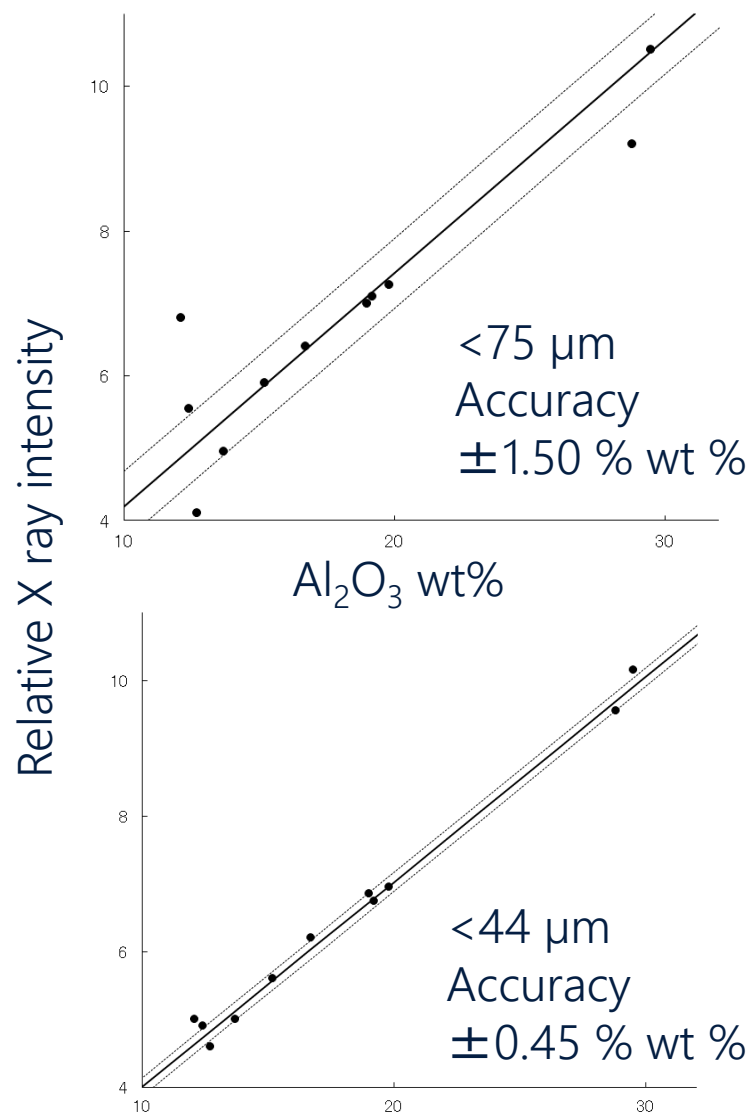
Standardset A



Standardset B



Al₂O₃ wt%



Observation

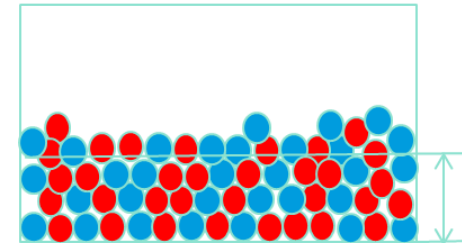
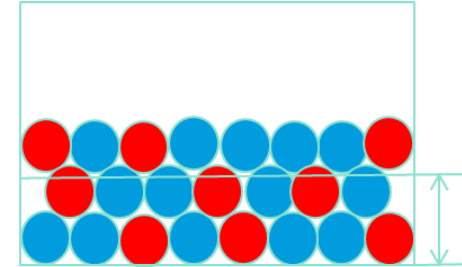
The dispersion of the calibration standards depends largely on the particle size


Recommendation for action

Optimizing the grind size can drastically improve the accuracy of a method

→ Shorter XRF measurement time possible

- The shape of the sample influences the measurement result
particle size (and distribution), preparation, etc.
- Method parameters include sample preparation
- Parameter optimization in sample preparation means recalibration
- Reproducibility of sample preparation is a priority





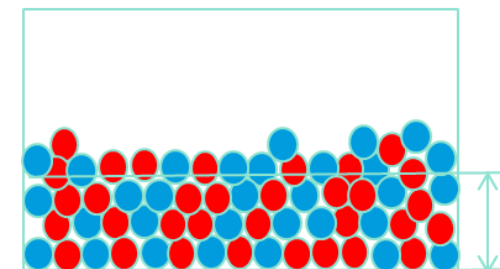
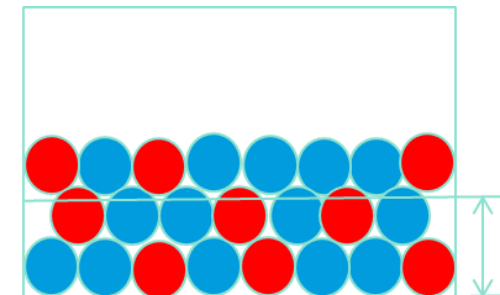
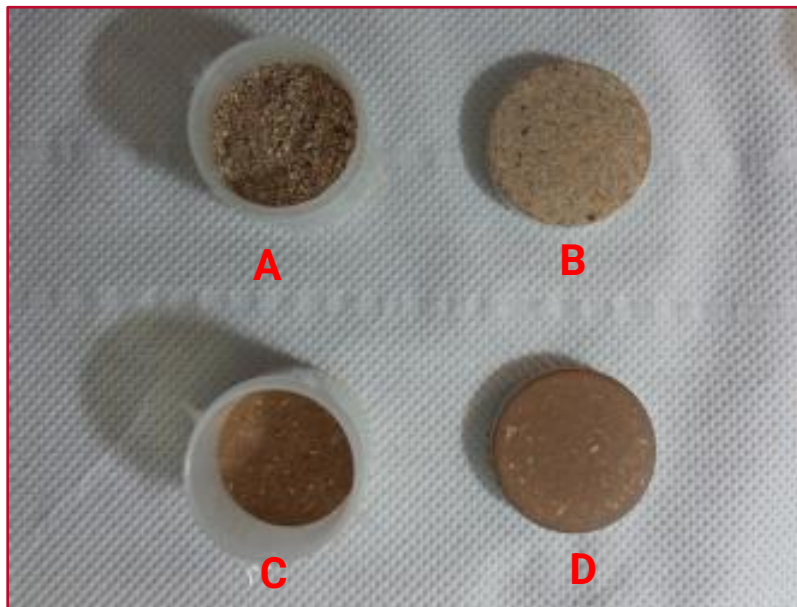
Influence of particle size on the XRF result

On the example of waste wood

SM 300



PM 400



(A) Powder pouring after 1st grinding (e.g. SM 300)

(B) Pellet from powder of the 1st grinding step

(C) Powder pouring after 2nd grinding (e.g. PM 400)

(D) pellet made of powder of the 2nd grinding step

Always dry biological samples well and store them in a dry place!

- Water absorbs and scatters
- Strength of pellets impaired

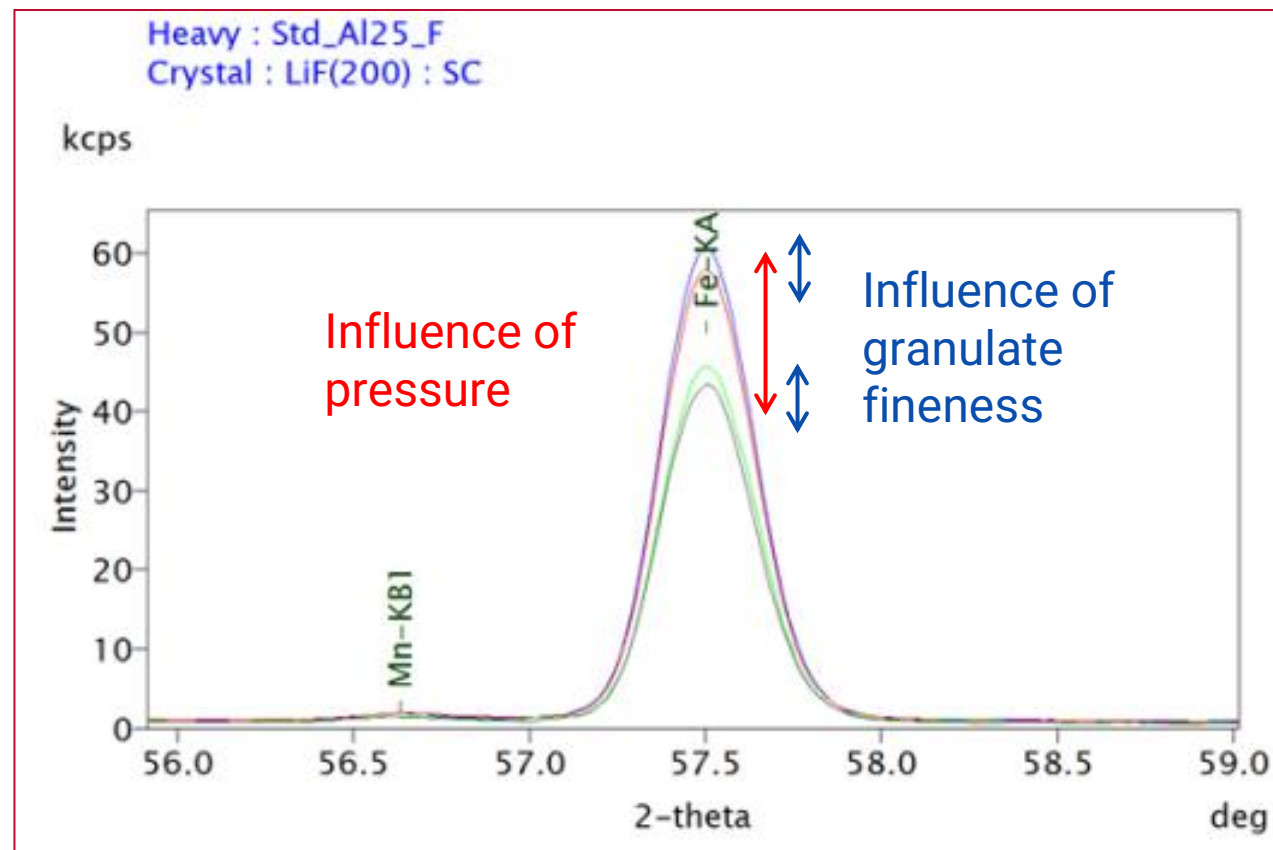
Heavy element (Fe)

(D) Pellet fine

(B) Pellet coarse

(C) Powder fine

(A) Powder coarse



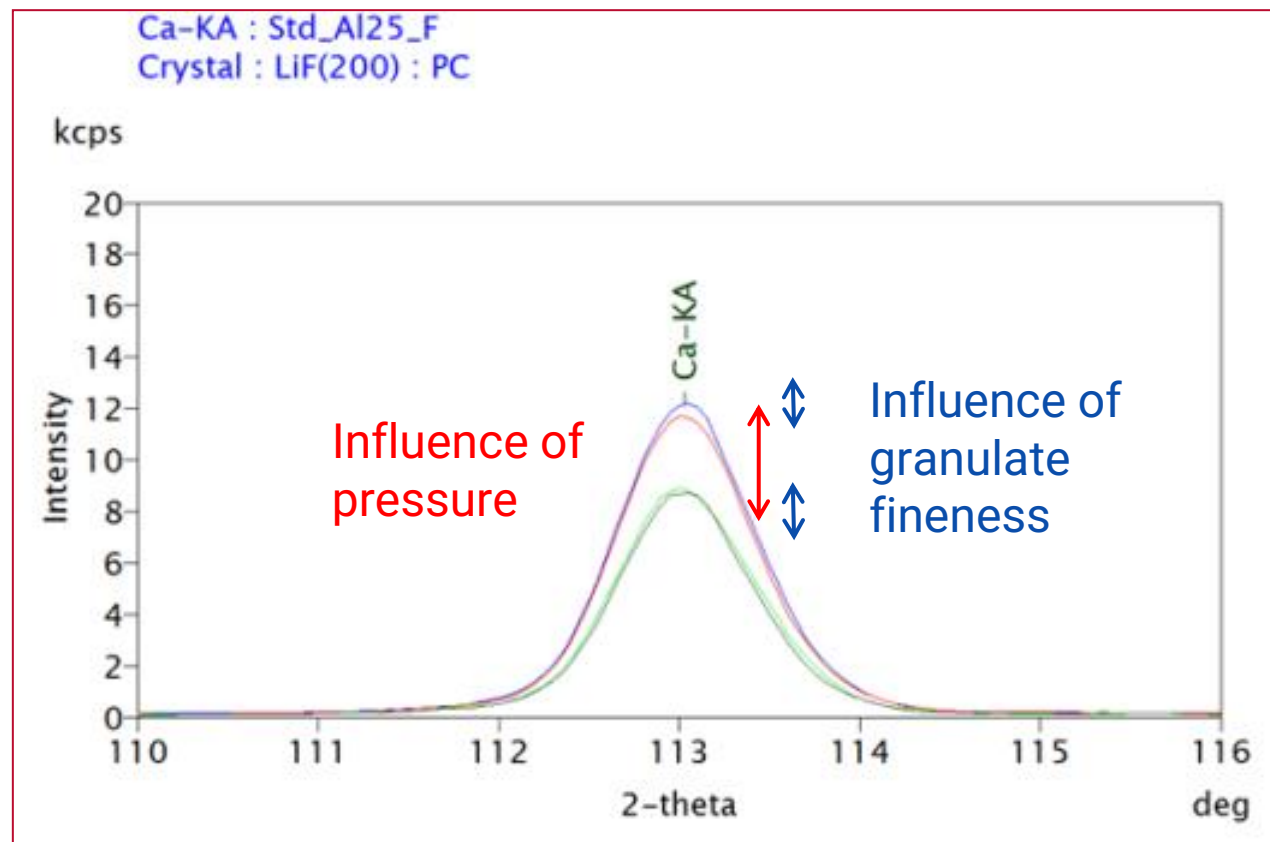
Medium element (Ca)

(D) Pellet fine

(B) Pellet coarse

(C) Powder fine

(A) Powder coarse



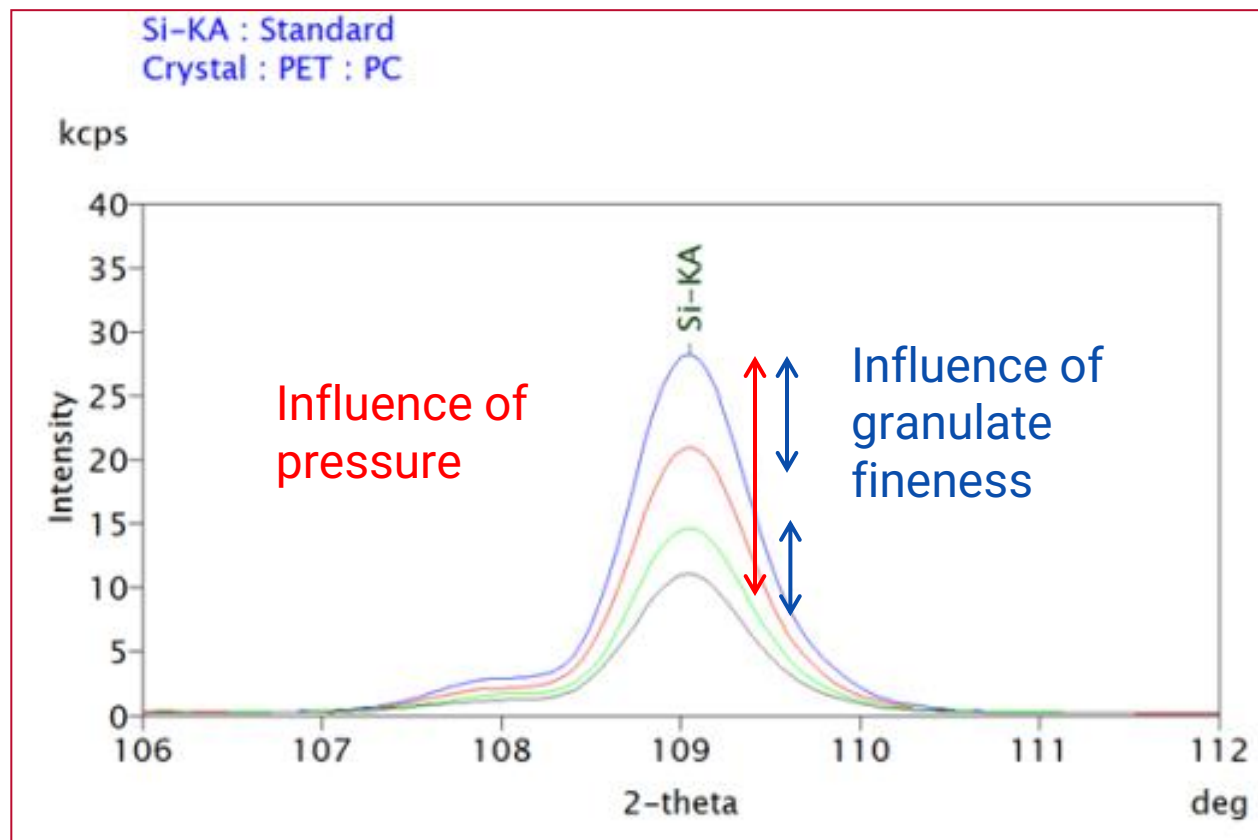
Light element (Si)

(D) Pellet fine

(B) Pellet coarse

(C) Powder fine

(A) Powder coarse



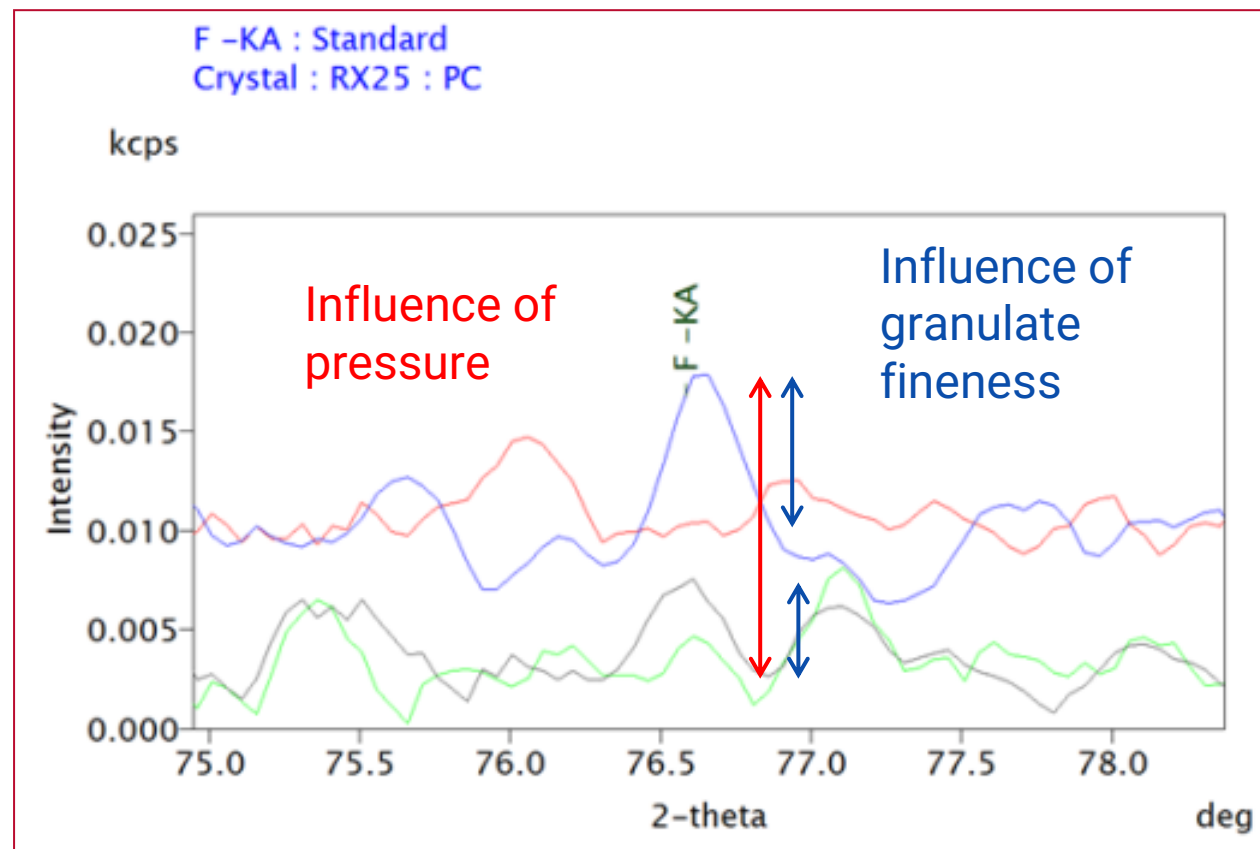
Very light element (F)

(D) Pellet fine

(B) Pellet coarse

(C) Powder fine

(A) Powder coarse





Signal intensity

- Fine powder & compact pellets, high signal intensity
- For lower detection limits, see F-analysis
- Reproducibility is a priority

Abrasion

- Finer powder = more intensive grinding = more abrasion
- Hard metal phases can reduce abrasion
- Which abrasion is the least interfering
→ Abrasion is unavoidable



Questions to the Influence of particle size?